

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH values and concentrations based on the pH and  $K_a$  or  $K_b$  and vice versa.
3. Calculate the pH of weak acids and bases.

## Vocabulary

- |                   |                             |                    |
|-------------------|-----------------------------|--------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid/base  |
| 4. autoionization | 5. acid rain                | 6. buffer solution |
| 7. autoionization | 8. acid rain                | 9. pH              |

## Key Equation

- |                                     |                                     |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

## Part A Completion

Write the equation and/or the numerical value(s) for the reaction(s) described. Show the calculation for pH or pOH. Write the chemical formula for the acid or base.

1. What is the pH of a 0.1 M solution of hydrochloric acid?
2. What is the pH of a 0.1 M solution of sodium hydroxide?
3. What is the pH of a 0.1 M solution of acetic acid?
4. What is the pH of a 0.1 M solution of ammonia?
5. What is the pH of a 0.1 M solution of hydrofluoric acid?
6. What is the pH of a 0.1 M solution of hydrocyanic acid?
7. What is the pH of a 0.1 M solution of hydrobromic acid?
8. What is the pH of a 0.1 M solution of hydroiodic acid?
9. What is the pH of a 0.1 M solution of hydrochloric acid?
10. What is the pH of a 0.1 M solution of hydrobromic acid?
11. What is the pH of a 0.1 M solution of hydroiodic acid?
12. What is the pH of a 0.1 M solution of hydrochloric acid?
13. What is the pH of a 0.1 M solution of hydrobromic acid?
14. What is the pH of a 0.1 M solution of hydroiodic acid?
15. What is the pH of a 0.1 M solution of hydrochloric acid?
16. What is the pH of a 0.1 M solution of hydrobromic acid?
17. What is the pH of a 0.1 M solution of hydroiodic acid?
18. What is the pH of a 0.1 M solution of hydrochloric acid?
19. What is the pH of a 0.1 M solution of hydrobromic acid?
20. What is the pH of a 0.1 M solution of hydroiodic acid?

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# 19 2 Hydrogen Ions And Acidity Work Sheet Answers

**CL Gary**



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