

- Which best describes the current atomic theory?
  - atoms consist of electrons circling in a definite orbits around a positive nucleus.
  - atoms are composed of electrons in a cloud around a positive nucleus.
  - Atoms can easily be split, at which time they become radioactive.
  - an atom's mass is determined by the mass of its neutrons.
- What is the nuclear composition of uranium-235?
  - 92 electrons + 143 protons
  - 92 protons + 143 electrons
  - 143 protons + 92 neutrons
  - 92 protons + 143 neutrons
- Which best describes the relationship between subatomic particles in any neutral atom?
  - the number of protons equals the number of electrons
  - the number of protons equals the number of neutrons
  - the number of neutrons equals the numbers of electrons
  - the number of neutrons is greater than the number of protons
- What is the name of the compound  $\text{PbO}_2$ ?
  - lead oxide
  - lead(II) oxide
  - lead oxide (I)
  - lead (IV) oxide
- What is the name of  $\text{HCl(aq)}$ ?
  - chloric acid
  - hydrochloric acid
  - hydrogen chloride
  - perchloric acid
- What is the formula for calcium nitrate?
  - $\text{CaNO}_2$
  - $\text{Ca}(\text{NO}_2)_2$
  - $\text{Ca}(\text{NO}_3)_2$
  - $\text{Ca}_2\text{N}_2$
- What is the correct formula for dinitrogen pentoxide?
  - $\text{N}_4\text{O}$
  - $\text{NO}_2$
  - $\text{N}_2\text{O}_5$
  - $\text{NO}_4$
- If the volume of 18.5 gram piece of metal is  $2.35 \text{ cm}^3$ , what is the identity of the metal?
  - iron
  - lead
  - nickel
  - zinc

9. Which substance listed in the table is a liquid at  $27^\circ\text{C}$ ?

a. I    b. II    c. III    d. IV

	Melting Point	Boiling Point
I	$28^\circ\text{C}$	$146^\circ\text{C}$
II	$-16^\circ\text{C}$	$28^\circ\text{C}$
III	$20^\circ\text{C}$	$146^\circ\text{C}$
IV	$-90^\circ\text{C}$	$14^\circ\text{C}$

- Which will increase the solubility of MOST solid solutes?
  - decreasing the temperature
  - decreasing the amount of solvent at constant temperature
  - increasing the amount of solute at constant temperature
  - increasing the temperature

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**S Marginson**



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