Structure of an Atom Guided Notes

Atoms are the building blocks of matter. They are the smallest particle into which an element can be divided and still retain its identity. An element is matter that is made up of only one type of atom.

Atoms have substructure of their own. They have a nucleus, which contains protons and neutrons. Protons have a positive charge and a mass of 1 amu (atomic mass unit). Neutrons have no charge and are

also measured at 1 amu. The number of protons in the atomic nucleus, which is the atomic number, is the defining characteristic of each element.

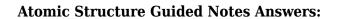
Electrons, which have a negative charge and a mass of almost zero, are located in an area surrounding the nucleus called the electron cloud. The way electrons react between charged particles explains the way atoms form molecules.

WORD BANK

Matter atomic number electrons molecules no charge positive charge divided element nucleus protons electron cloud neutrons element

Atomic Structure Guided Notes Answers

N Noddings



Whispering the Secrets of Language: An Psychological Journey through Atomic Structure Guided Notes Answers

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