









Matter

STEM

Matter

Atoms – Elements – Molecules – Compounds

Atoms	Elements	Molecules	Compounds
<p>The smallest unit of matter and has three main parts (protons, neutrons, and electrons). The basic building blocks of matter. Smallest part of an element.</p> <div style="text-align: center;"> </div> <p style="text-align: center;">Example Oxygen, Hydrogen, Nitrogen</p>	<p>A pure substance consisting of one type of atom and can be identified by its atomic number (number of protons). All elements are found on the Periodic Table of Elements and have a chemical symbol.</p> <div style="text-align: center;"></div> <p style="text-align: center;">Example Oxygen's atomic number is 8</p>	<p>Formed when two or more atoms join together chemically. Can be the same atoms.</p> <div style="text-align: center;"> or  or </div> <p style="text-align: center;">Example Oxygen, Water, Nitrogen, Carbon Dioxide</p>	<p>Formed when two different elements join together. All compounds are molecules, but all molecules are not compounds.</p> <div style="text-align: center;"> or </div> <p style="text-align: center;">Example Water, Carbon Dioxide</p>

Atoms, Elements, Molecules and Compounds

Atoms Molecules Elements Matter Energy

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