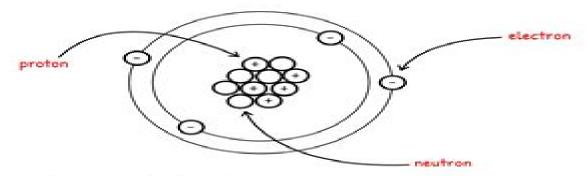
Name	answer key	 Date	
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Atomic Structure

Label the parts of an atom on the diagram below, and answer the questions!



- 1. What type of charge does a proton have? Positive (+1)
- 2. What type of charge does a neutron have?

 Neutral (0)
- 3. What type of charge does an electron have? Negative (-1)
- 4. Which two subatomic particles are located in the nucleus of an atom? Protons and neutrons
- 5. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus? 35
- 6. What is the atomic number of the atom in the diagram above?
- 7. What is the atomic mass/mass number of the atom in the diagram above? (11+11=22) "protons and neutrons added together) 22 is the atomic mass/mass number
- 8. How many protons are in the nucleus of an atom with an atomic number of 15?

 15 "number of protons is the same as the atomic number."
- How many electrons are in the nucleus of an atom with an atomic number of 20?
- How many neutrons are in the nucleus of an atom with an atomic number of 25?
 (use Periodic Table for mass) 30 Neutrons, atomic mass; 54,938~ 55 atomic number; 25....55-25-30
- 11. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?

 Hass number = 6.94 (Li-Lithium has an atomic number of 3 which means it has 3 protons.
- 12. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?

Neutrons = 11 (atomic mass: 36 - 25:atomic number)

Atomic Structure Guided Practice Problems Answers

Arshad Iqbal

Atomic Structure Guided Practice Problems Answers:

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