More aqueous equilibrium self test/practice problems, Dr. Scott Ensign.

 (Acid-base equilibrium review) A solution is prepared by dissolving 0.15 moles of acetic acid in 1.00 liters of solution. K<sub>a</sub> for acetic acid is 1.8 x 10<sup>-5</sup>. What will be the pH of the solution once equilibrium is established?

a. 2.0 b. 5.6

c. 2.8 d. 3.9

e. 4.7

2. (Acid-base equilibrium review) A solution is prepared by dissolving 0.10 moles of sodium acetate (NaCH<sub>3</sub>COO) in 1.00 liters of solution. K<sub>a</sub> for acetic acid is 1.8 x 10<sup>-3</sup>. What will be the pH of the solution once equilibrium is established?

a. 8.9 b. 5.1

c. 9.9

d. 4.1 e. 9.3

3. A solution is prepared by mixing 0.15 moles of acetic acid (CH<sub>2</sub>COOH) with 0.10 moles of sodium acetate (NaCH<sub>2</sub>COO) in 1.00 liters of solution. K<sub>a</sub> for acetic acid is 1.8 x 10<sup>-5</sup>. What will be the pH of the solution once equilibrium is established?

a. 4.9 b. 6.6

c. 2.9

d. 4.6

e. 4.7

4. A solution is prepared by mixing 0.15 moles of acetic acid (CH<sub>2</sub>COOH) with 0.050 moles of nitric acid (HNO<sub>2</sub>) in 1.00 liters of solution. K<sub>a</sub> for acetic acid is 1.8 x 10<sup>-5</sup>. What will be the pH of the solution once equilibrium is established?

a. 2.4 b. 2.8

c. 1.3

d. 4.3

e. 5.2

5. (Acid-base equilibrium review) What would be the pH of a solution prepared by dissolving 0.55 moles of NH<sub>3</sub> in 1.00 liters of solution?  $K_b$  for NH<sub>3</sub> is 1.80 x 10<sup>-5</sup>

a. 9.51 b. 11.5

c. 11.8

d. 2.50

e. 9.00

# **Aqueous Equilibrium Practice Problems**

Kenneth W. Raymond

### **Aqueous Equilibrium Practice Problems:**

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Unveiling the Magic of Words: A Review of "Aqueous Equilibrium Practice Problems"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "Aqueous Equilibrium Practice Problems," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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