

Problems

(References for these problems will be found on page 1271.)

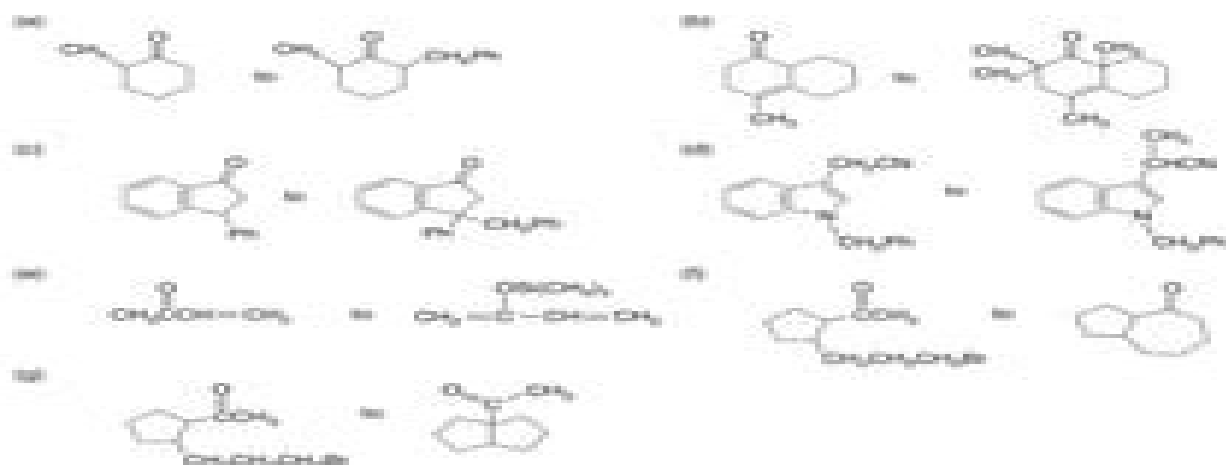
1.1. Arrange each series of compounds in order of decreasing acidity.

- (a) CH3CH2NO2, CH3CH2C(=O)CH3, CH3CH2C(=O)CH2CH3, CH3CH2CH2CH3
 (b) CH3CH2NH2, CH3CH2OH, CH3CH2CH2CH3, CH3CH2CH2CH2CH3
 (c) CH3COCH2CH2CH3, CH3COCH2CH2CH2CH3, CH3COCH2CH2CH2CH2CH3, CH3COCH2CH2CH2CH2CH2CH3
 (d) HOCH2CH2CH2CH2CH2CH2CH2CH2OH, HOCH2CH2CH2CH2CH2CH2CH2CH2CH2OH, HOCH2CH2CH2CH2CH2CH2CH2CH2CH2CH2OH, HOCH2CH2CH2CH2CH2CH2CH2CH2CH2CH2CH2OH

1.2. Write the structures of all possible isolates for each ketone. Indicate which you expect to be favored in a kinetically controlled deprotonation. Indicate which you would expect to be the most stable isolate.



1.3. Suggest reagents and reaction conditions that would be suitable for effecting each of the following conversions.



Advanced Organic Chemistry Part A Solution

RM Cervero

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Advanced Organic Chemistry Part A Solution:

Advanced Organic Chemistry Part A Solution Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Advanced Organic Chemistry Part A Solution**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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