

Advanced Theory practice Test

Multiple choice (single answer) to the three main components of the question.

1. (1.1.1.1) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By calculating the number of protons/neutrons
 - By identifying the number of neutrons

2. (1.1.1.2) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By identifying the number of protons/neutrons
 - By identifying the number of neutrons

3. (1.1.1.3) Which of the following is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By identifying the number of protons/neutrons
 - By identifying the number of neutrons



4. (1.1.1.4) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By identifying the number of protons/neutrons
 - By identifying the number of neutrons

5. (1.1.1.5) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By identifying the number of protons/neutrons
 - By identifying the number of neutrons

Figure 1:



- (a) 14 protons, 14 electrons, 14 neutrons
(b) 14 protons, 14 electrons, 14 neutrons
(c) 14 protons, 14 electrons, 14 neutrons

6. (1.1.1.6) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By identifying the number of protons/neutrons
 - By identifying the number of neutrons

- (a) 14 protons, 14 electrons, 14 neutrons
(b) 14 protons, 14 electrons, 14 neutrons

Atomic Theory Practice Test

Neville G. Warren

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Atomic Theory Practice Test Book Review: Unveiling the Magic of Language

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