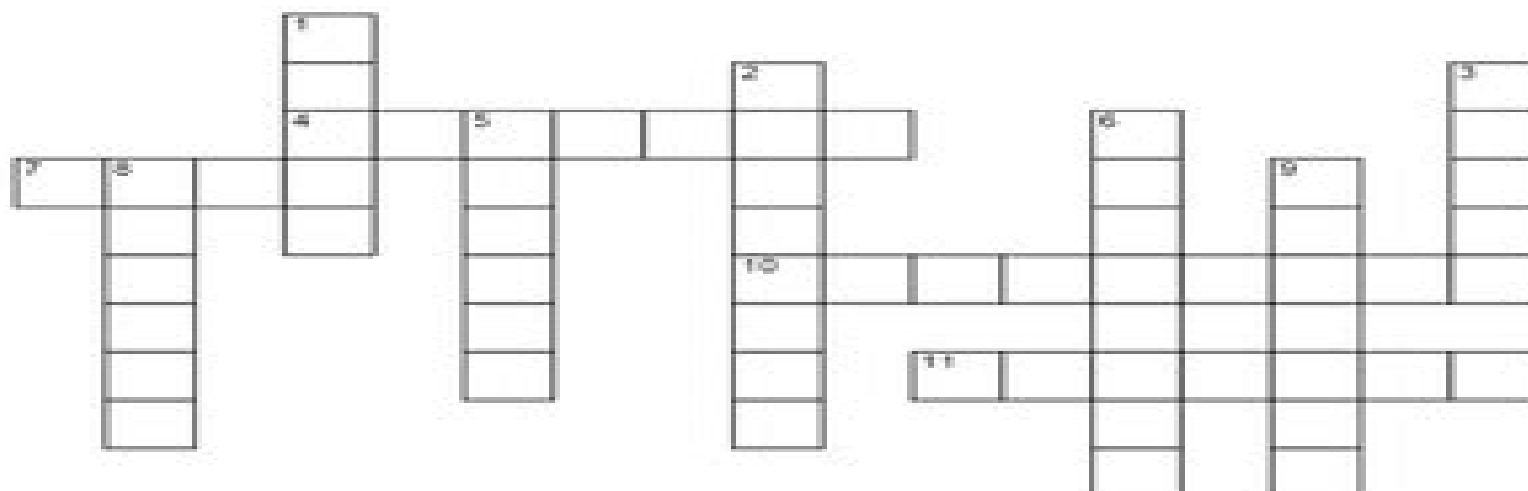


## Atomic structure linkword



### CLUES

- 9 down An atom is composed of equal numbers of \_\_\_\_\_ and electrons. (7)
- 10 across An ion has a different number of protons and \_\_\_\_\_, and so it is charged. (9)
- 8 down Proton number is also called "\_\_\_\_\_ number" and has the symbol  $Z$ . (6)
- 7 across \_\_\_\_\_ number is also called "nucleon number" and has the symbol  $A$ . (4)
- 3 down The number of neutrons in each isotope is given by  $A$  \_\_\_\_\_  $Z$ . (5)
- 6 down Two versions of the same element will have different masses and different numbers of \_\_\_\_\_, but the same number of protons. (8)
- 11 across Each version of an element with the same proton number but different mass numbers is known as an \_\_\_\_\_. (7)
- 4 across Electrons occupy the space around the \_\_\_\_\_. (7)
- 5 down The nucleus lies at the \_\_\_\_\_ of the atom and contains only protons and neutrons. (5)
- 2 down Protons and neutrons collectively are sometimes called \_\_\_\_\_. (8)
- 1 down The nucleus is tiny, positive and \_\_\_\_\_. (5)

**WORD BANK:** PROTONS, NUCLEUS, NUCLEONS, NEUTRONS, MINUS, MASS, ISOTOPE, ELECTRONS, DENSE, center, ATOMIC

# Atomic Structure Quick Xword

**Clemens Wendtner**



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