

Balancing Chemical Equations Worksheet

1. 2 H<sub>2</sub> + 1 O<sub>2</sub> → 2 H<sub>2</sub>O
2. 1 N<sub>2</sub> + 3 H<sub>2</sub> → 2 NH<sub>3</sub>
3. 1 S<sub>8</sub> + 12 O<sub>2</sub> → 8 SO<sub>3</sub>
4. 1 N<sub>2</sub> + 1 O<sub>2</sub> → 2 N<sub>2</sub>O
5. 2 HgO → 2 Hg + 1 O<sub>2</sub>
6. 6 CO<sub>2</sub> + 6 H<sub>2</sub>O → 1 C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + 6 O<sub>2</sub>
7. 1 Zn + 2 HCl → 1 ZnCl<sub>2</sub> + 1 H<sub>2</sub>
8. 1 SiCl<sub>4</sub> + 4 H<sub>2</sub>O → 1 H<sub>4</sub>SiO<sub>4</sub> + 4 HCl
9. 2 Na + 1 H<sub>2</sub>O → 2 NaOH + 1 H<sub>2</sub>
10. 2 H<sub>3</sub>PO<sub>4</sub> → 1 H<sub>4</sub>P<sub>2</sub>O<sub>7</sub> + 1 H<sub>2</sub>O
11. 1 C<sub>10</sub>H<sub>16</sub> + 8 Cl<sub>2</sub> → 10 C + 16 HCl
12. 1 CO<sub>2</sub> + 2 NH<sub>3</sub> → 1 OC(NH<sub>2</sub>)<sub>2</sub> + 1 H<sub>2</sub>O
13. 4 Si<sub>2</sub>H<sub>3</sub> + 17 O<sub>2</sub> → 8 SiO<sub>2</sub> + 6 H<sub>2</sub>O<sub>3</sub>
14. 2 Al(OH)<sub>3</sub> + 3 H<sub>2</sub>SO<sub>4</sub> → 1 Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + 6 H<sub>2</sub>O
15. 4 Fe + 3 O<sub>2</sub> → 2 Fe<sub>2</sub>O<sub>3</sub>
16. 1 Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + 6 KOH → 3 K<sub>2</sub>SO<sub>4</sub> + 2 Fe(OH)<sub>3</sub>
17. 2 C<sub>7</sub>H<sub>6</sub>O<sub>2</sub> + 15 O<sub>2</sub> → 14 CO<sub>2</sub> + 6 H<sub>2</sub>O
18. 1 H<sub>2</sub>SO<sub>4</sub> + 8 HI → 1 H<sub>2</sub>S + 4 I<sub>2</sub> + 4 H<sub>2</sub>O
19. 4 FeS<sub>2</sub> + 11 O<sub>2</sub> → 2 Fe<sub>2</sub>O<sub>3</sub> + 8 SO<sub>2</sub>
20. 2 Al + 3 FeO → 1 Al<sub>2</sub>O<sub>3</sub> + 3 Fe
21. 1 Fe<sub>2</sub>O<sub>3</sub> + 3 H<sub>2</sub> → 2 Fe + 3 H<sub>2</sub>O
22. 1 Na<sub>2</sub>CO<sub>3</sub> + 2 HCl → 2 NaCl + 1 H<sub>2</sub>O + 1 CO<sub>2</sub>
23. 2 K + 1 Br<sub>2</sub> → 2 KBr
24. 1 C<sub>7</sub>H<sub>16</sub> + 11 O<sub>2</sub> → 7 CO<sub>2</sub> + 8 H<sub>2</sub>O
25. 1 P<sub>4</sub> + 5 O<sub>2</sub> → 2 P<sub>2</sub>O<sub>5</sub>

# Balancing Chemical Equations Answers Cavalcade

**Crispin Collins**

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