

a		<p>Describe the process of spermatogenesis leading to the production of four sperm cells in a human male.</p> <ul style="list-style-type: none"> <li>a. in the seminiferous tubule ✓</li> <li>b. diploid cells/spermatogonia grow/enlarge ✓</li> <li>c. two divisions of meiosis ✓</li> <li>d. primary spermatocyte carries out the first division and secondary spermatocytes carry out the second division ✓</li> <li>e. meiosis produces haploid cells/spermatids ✓</li> <li>f. haploid cells/spermatids <u>differentiate</u> into spermatozoa/sperm cells OR develop tail/flagellum/helical mitochondrion/acrosome/sac of enzymes ✓</li> <li>g. Sertoli cells help sperm to mature/differentiate</li> </ul>	<p>Allow answers presented in a clearly annotated diagram.</p>	4 max
b		<p>Outline the roles of estrogen and progesterone in females during human reproduction.</p> <ul style="list-style-type: none"> <li>a. estrogen/progesterone for repair/thickening/development/vascularisation of uterus lining/endometrium ✓</li> <li>b. estrogen (at high levels) stimulates LH secretion (which stimulates ovulation)</li> <li>c. progesterone maintains the uterus lining/endometrium during pregnancy/prevents miscarriage ✓</li> <li>d. progesterone/estrogen inhibits FSH/LH secretion ✓</li> <li>e. progesterone for development of breast tissue during pregnancy ✓</li> <li>f. fall in progesterone/rise in estrogen leads to labour/contractions/childbirth ✓</li> <li>g. estrogen/progesterone cause pre-natal development of female reproductive organs OR</li> <li>h. estrogen/progesterone cause secondary sexual characteristics ✓</li> </ul>		4 max

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chemical equation  $2\text{Fe}_2\text{O}_3 + 3\text{C} \rightarrow 4\text{Fe} + 3\text{CO}_2$  on towards the next step now step 3 once you are done with this check out if the number

of atoms for the other elements is also equal on both the sides

what is stoichiometry formula example balancing equations - Jan 09 2023

web 1 78 348 stoichiometry helps us determine how much substance is needed or is present things that can be measured are

reactants and products mass molecular weight chemical equations formulas stoichiometric coefficient the stoichiometric

coefficient or stoichiometric number is the number of molecules that participate in the reaction

solved balancing equations and simple stoichiometry balance - Sep 05 2022

web expert answer transcribed image text balancing equations and simple stoichiometry balance the following equations  
 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$   
 $2\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow 3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr} + \text{NaBr} + \text{SNO}$   
 $4\text{NaSO}_3 + \text{Ga} + \text{SO}_3 \rightarrow 3\text{NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$  using the equation from problem 2 above answer the following

5 3 stoichiometry calculations chemistry libretexts - Mar 11 2023

web learning objectives to balance equations that describe reactions in solution to calculate the quantities of compounds produced or consumed in a chemical reaction to solve quantitative problems involving the stoichiometry of reactions in solution

*stoichiometry practice problems chemistry steps* - Jun 02 2022

web stoichiometry stoichiometry practice problems this is a comprehensive end of chapter set of practice problems on stoichiometry that covers balancing chemical equations mole ratio calculations limiting reactants and percent yield concepts the links to the corresponding topics are given below

**balancing equations and simple stoichiometry key solano** - Aug 16 2023

web balancing equations and simple stoichiometry key balance the following equations  
 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$   
 $2\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   
 $3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + 2\text{CO}_2 + 3\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{SO}_3 + 6\text{NaBr} + 3\text{SNO} + 2\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$  using the following equation  
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

*ideal stoichiometry practice khan academy* - Jan 29 2022

web ideal stoichiometry google classroom you might need calculator periodic table given the following reaction  
 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_x$   
 $2\text{ZnCl}_x + 2\text{Cu}$  how many moles of  $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_x$  will be produced from 23.0 g of  $\text{Ce} + \text{Zn}$  assuming  $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_x$  is available in excess

balancing equations and simple stoichiometry key yumpu - Nov 07 2022

web jul 22 2013 balance the following equations  
 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$   
 $2\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   
 $3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + 2\text{CO}_2 + 3\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{SO}_3 + 6\text{NaBr} + 3\text{SNO} + 2\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$  using the following equation  
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

**3 stoichiometry chemical formulas and equations** - May 01 2022

web the stoichiometry of a balanced chemical equation identifies the maximum amount of product that can be obtained the stoichiometry of a reaction describes the relative amounts of reactants and products in a balanced chemical equation

solved stoichiometry practice worksheet balancing equations chegg - Feb 27 2022

web question stoichiometry practice worksheet balancing equations and simple stoichiometry balance the following equations  
 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$   
 $2\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   
 $3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr} + \text{Na}_2\text{SO}_3 + \text{Ga} + \text{SO}_3 + 3\text{NaBr} + \text{SNO} + \text{NF}_3$



snf2

[5 2 1 practice problems reaction stoichiometry](#) - Oct 06 2022

web may 28 2020 write the balanced equation for the reaction and calculate how much  $\text{SiO}_2$  is required to produce 3.00 kg of  $\text{SiC}$  answer  $\text{Ce SiO}_2 \rightarrow \text{SiC}$

**solved stoichiometry practice worksheet balancing equations** - Jul 03 2022

web question stoichiometry practice worksheet balancing equations and simple stoichiometry balance the following equations  $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$   $\text{C}_6\text{H}_{10}\text{O}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{KBr} + \text{GaBr}_3 \rightarrow \text{GaBr}_4$   $\text{Na}_2\text{SO}_3 + \text{O}_2 \rightarrow \text{Na}_2\text{SO}_4$   $\text{Sn} + \text{F}_2 \rightarrow \text{SnF}_4$   $\text{N}_2 + \text{O}_2 \rightarrow \text{NO}_2$   $\text{CO}_2 + \text{NaBr} \rightarrow \text{Na}_2\text{CO}_3 + \text{NaBr}$  solve the following stoichiometry grams problems 6

*stoichiometry and balancing reactions chemistry libretexts* - Jun 14 2023

web jun 30 2023 stoichiometry and balanced equations in stoichiometry balanced equations make it possible to compare different elements through the stoichiometric factor discussed earlier this is the mole ratio between two factors in a chemical reaction found through the ratio of stoichiometric coefficients

*solved name unit 6 balancing equations and simple chegg* - Mar 31 2022

web chemistry chemistry questions and answers name unit 6 balancing equations and simple stoichiometry this worksheet is worth 10 points for problems 5-7 make sure to show your work using the factor label method please circle or highlight your answers

*balancing chemical equations how to walkthrough video khan academy* - Aug 04 2022

web gabrielle m 9 years ago i'm working on chemical reactions double and single replacement on flvs now my first question for this video is how do you have 4 aluminum atoms when it says  $2\text{Al}$  subscript 2 do you just add the 2 and the subscript 2 same thing goes with the  $\text{O}_3$  i am so lost

**balancing equations and simple stoichiometry pdf scribd** - Dec 08 2022

web balancing equations and simple stoichiometry answers balance the following equations  $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$   $\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$   $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 3\text{S} + 3\text{H}_2\text{O}$   $2\text{NF}_3 + 3\text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$

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web google classroom balance the following chemical equation  $\text{Mg} + \text{OH}^- \rightarrow \text{Mg}(\text{OH})_2$   $\text{HCl} + \text{MgCl}_2 \rightarrow \text{H}_2\text{O}$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem 7 4 1 x x y y theta  $\theta$  pi  $\pi$  8 5 2 0 9 6

[stoichiometry article chemical reactions khan academy](#) - Jul 15 2023

web a common type of stoichiometric relationship is the mole ratio which relates the amounts in moles of any two substances in a chemical reaction we can write a mole ratio for a pair of substances by looking at the coefficients in front of each species

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*chemical reactions and stoichiometry chemistry library khan academy* - Apr 12 2023

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