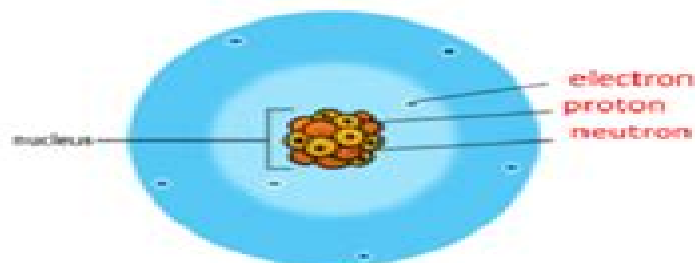


CHEMISTRY TEST STUDY GUIDE ANSWER KEY

Complete the following information about atoms.
Label the subatomic particles in the atom diagram below.



What charge does each subatomic particle have?

Proton: **positive**

Electron: **negative**

Neutron: **neutral (no charge)**

What subatomic particles are found in the nucleus?

protons and neutrons

Classify each of the following substances as an Element (E), Compound (C), or Mixture (M).

If it's a mixture, is it heterogeneous (he) or homogeneous (ho)?

- water **C**
- salt **C**
- air **M (ho)**
- carbon **E**
- sugar **C**
- saltwater **M (ho)**
- hydrogen **E**
- carbon dioxide **C**
- kool aid **M (ho)**

Use a periodic table to answer the following questions.

- Is sodium a metal or nonmetal? **metal**
- What is the atomic number of oxygen? **8**
- What is element symbol for iron? **Fe**
- What other elements are in the same family as Chlorine? **Fluorine, Bromine, Iodine, Astatine, Tennessine**
- Li, Be, B, C, N, O, F, and Ne are all in the same: **Period**

Adv Chemistry Study Guide Answer Key

Percy Gaillemard L. Smith



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Chemistry for Advanced Level Course Ted Lister, Janet Renshaw, 2000 This course study guide is designed to complement New Understanding Chemistry for Advanced Level but it can be used with any other core textbook for AS and A Level Chemistry as well. It aims to help further develop chemistry skills such as laboratory techniques, mathematical methods and data handling. The course study guide also provides outline solutions to a selection of questions and gives advice on answering all types of examination questions and support for Key Skills.

Advanced Chemistry for You Lawrie Ryan, 2000 Advanced Chemistry for You has been carefully designed to be interesting and motivating to the AS/A2 student with features that make it highly supportive of individual learning. Written by an experienced author team with the same straightforward approach as the successful New for You GCSE series.

New Understanding Chemistry for Advanced Level Third Edition Ted Lister, Janet Renshaw, 2000 Matches the specifications of the Awarding Bodies AQA, NEAB, AEB, OCR and Edexcel. This accessible text includes frequent hints, questions and examination questions providing support and facilitating study at home. It features photographs and comprehensive illustrations with 3D chemical structures.

Maths for Advanced Chemistry Michael Robinson, Mike Taylor, 2002 Written by teachers, this text is useful for both classroom work and homework exercises.

A-Level Chemistry Eileen Ramsden, Ray Barker, Darrel Barsby, Rob Commons, Gez Rizzo, Michala Swales, Ian Wood, J F Rounce, T L Lowe, Joan Sybil Chambers, D J Crawshaw, Brian Jefferson, David Bowles, Eddie Mullan, Garry Wiseman, John Rayneau, Mike Heylings, Rob Wagner, Steve Cavill, Tony Beadsworth, C P Rourke, Mark Gaulter, Brian Gaulter, Robert Smedley, Ian Cook, 2020-10-08 This highly regarded textbook covers all the main A Level Chemistry specifications.

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