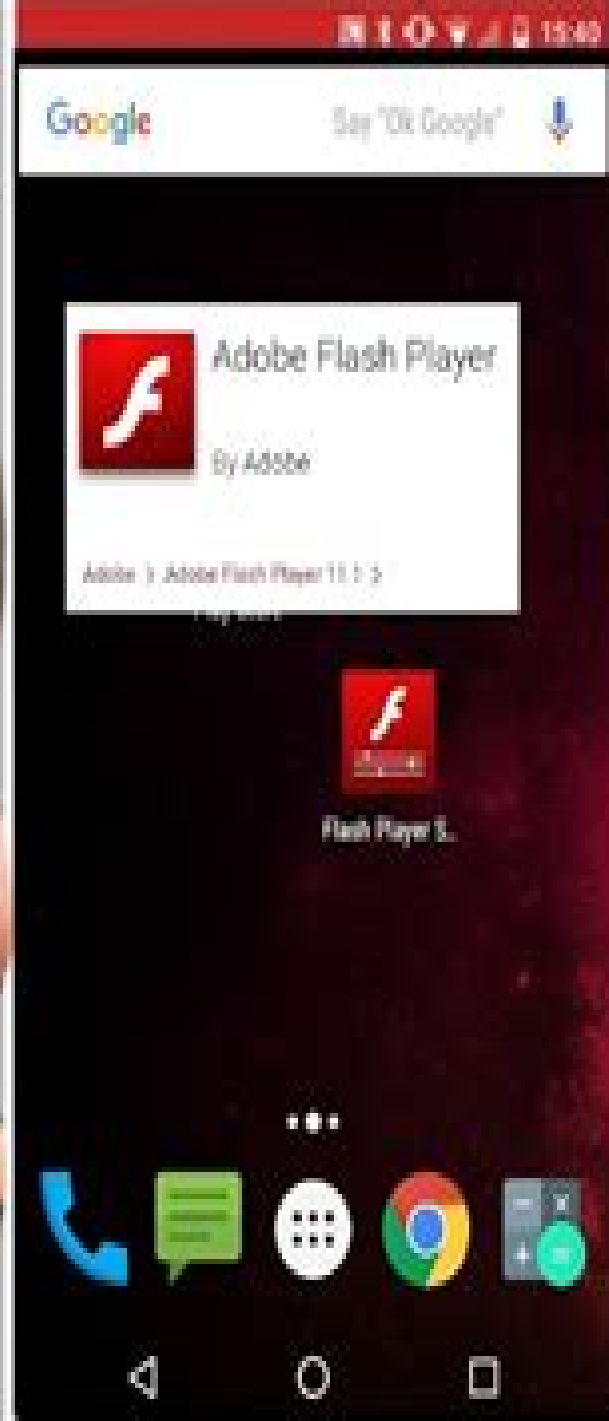




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**Proceedings of the Fifth International Conference on Arts and Design Education (ICADE 2022)** Juju Masunah, Tati Narawati, Yudi Sukmayadi, Zakarias S. Soeteja, Trianti Nugraheni, Rita Milyartini, Agus Budiman, 2023-09-27 This is an open access book The 5th INTERNATIONAL CONFERENCE ON ARTS AND DESIGN EDUCATION ICADE Digital Media Literacy and Mastery of Post pandemic in Arts and Design Education BACKGROUND The past year has been one of enormous turbulence and upheaval where nobody has been left untouched by the impact of the global pandemic even a great change has been forced upon artists and educators in their work creation process as well as learning and teaching practice What and how artists and educators have thought and gone through to keep their field sustained should be disseminated as a mutual learning to enhance our awareness on arts and design sustainability and role during pandemic ICADE 5 aims to bring together leading academic artists arts and design researchers and scholars to exchange and share their notion experiences

and research results on all aspects of Digital Media Literacy and Mastery in Arts and Design education during pandemic and its urgencies after pandemics It also provides a premier interdisciplinary platform for arts and design researchers practitioners and educators to present and discuss the most recent innovations trends and concerns as well as practical challenges encountered and solutions adopted which involves digital media literacy or mastery in the fields of Arts and Design education The 5th International Conference on Arts and Design Education ICADE is a conference hosted by Faculty of Arts and Design Education Universitas Pendidikan Indonesia This conference will be held on September 22nd 2022 through Zoom Webinar ICADE has covered a broad range of topic on music dance drama visual arts visual communication design arts education film and television and multidisciplinary arts fields This conference scopes may pertain how artist designer or arts and design educators finally adapt and conquer digital media to support their work creation and studies activities during pandemic attitude changes toward art and design works creation best practice of art and design creation and studies during pandemic sorts of digital media which might be left or continuously used after pandemic what finally the digital media can t ever cover during pandemic even to this moment post pandemic potential disruptive constructive point of digital media use as well as other substance where we spot a relation between a digital media with arts and design creation and studies

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web isotopes are atoms of the same element that differ in the amount of neutrons and atomic mass almost all elements on the periodic table have at least two different natural isotopes many elements

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web the isotopes of a given element always contain the same number of protons and therefore occupy the same place on the periodic table however because isotopes contain different numbers of neutrons each isotope has a unique atomic mass

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web apr 18 2015 isotopes are variants of a particular chemical element which differ in neutron number although all isotopes of a given element have the same number of protons in each atom answer link

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web jul 25 2014 isotopes differ in the number of neutrons in ions the number of electrons is different from the number of protons isotopes are atoms that have the same number of protons but different numbers of neutrons thus atoms of  $^{12}_6\text{C}$  and of  $^{13}_6\text{C}$  are isotopes of each other

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web isotopes are those atoms having same atomic number number of protons are same but different mass number number of neutrons differ ions are atoms which contain an overall charge where number of protons number of electrons

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web atoms of the same element i e same atomic number  $Z$  that have different numbers of neutrons are called isotopes for example 99 of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1 of the carbon atoms have 7

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web aug 9 2023 how do different isotopes of an element differ two isotopes differ in the number of neutrons in the nucleus for example carbon 14 has 8 neutrons differing from carbon 12 which has 6

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web atoms of the same element that differ in their numbers of neutrons are called isotopes many isotopes occur naturally usually one or two isotopes of an element are the most stable and common different isotopes of an element generally have the same physical and chemical properties because they have the same numbers of protons and electrons

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web jan 5 2016 explanation isotopes are atoms of the same element same atomic number i.e. same number of protons in the nucleus but different atomic masses different mass numbers due to different numbers of neutrons in the nucleus for example we may get an isotope of chlorine with 17 protons and 18 neutrons hence atomic mass 35

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web may 16 2018 explanation an atom is characterized as a particular element by the number of positively charged massive nuclear particles i.e.  $Z$   $Z=1$  we got hydrogen  $Z=2$  we got helium  $Z=3$  lithium  $Z=4$  we got silver the nucleus can also contain neutrons massive nuclear particles of zero charge

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web atoms of the same element with different numbers of neutrons are called isotopes close isotope atoms of an element with the same number of protons and electrons but different numbers of neutrons

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web oct 31 2023 an isotope is one of two or more species of atoms of a chemical element with the same atomic number and position in the periodic table and nearly identical chemical behavior but with different atomic masses and physical properties

every chemical element has one or more isotopes

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web jul 18 2018 answer link due to the number of neutrons isotopes differ with respect to the number of neutrons their nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus d more than one response is correct e they have different numbers of protons in the nucleus