

# Study Guide Atomic Structure and Periodic Table

Circle the possible best answer for each question: (1pt each)

Neutrons are subatomic particles that are located

- a. In the nucleus and have a positive charge
- b. In the nucleus and have a negative charge
- c. In the nucleus and have no charge
- d. Around the nucleus and have a positive charge

The atomic number of an element is equal to the number of

- a. Electrons
- b. Neutrons
- c. Protons
- d. Protons plus electrons

Which of the pair below is equal to each other in an atom

- a. Protons and electrons
- b. Neutrons and protons
- c. Neutrons and electrons
- d. Mass number and electrons

In the late 1800s, experiments using cathode ray tubes led to the discovery of the

- a. Positron
- b. Neutron
- c. Proton
- d. Electron

Which two particles each have a mass approximately equal to one atomic mass unit?

- a. Electrons and positron
- b. Proton and electron
- c. Electron and neutron
- d. Proton and neutron

To which group of the periodic tables do lithium and potassium belong?

- a) Alkali metals
- b) Alkaline earth metals
- c) transition metals
- d) halogens

To which group of the periodic tables do Fluorine and Chlorine belong?

- a) Alkali metals
- b) Alkaline Earth metals
- c) Transition metals
- d) halogens

# Atomic Structure And Periodic Relationships Study Guide

**Neil D. Jespersen, Pamela Kerrigan**



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**A Student's Guide Through the Great Physics Texts** Kerry Kuehn,2015-12-17 This book provides a chronological introduction to modern atomic theory which represented an attempt to reconcile the ancient doctrine of atomism with careful experiments performed during the 19th century on the flow of heat through substances and across empty space Included herein are selections from classic texts such as Carnot s Reflection on the Motive Power of Fire Clausius Mechanical Theory of Heat Rutherford s Nuclear Constitution of Atoms Planck s Atomic Theory of Matter and

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Catalogue of the University of Michigan University of Michigan, 1965 Announcements for the following year included in some vols **Announcement** University of Michigan. College of Engineering, 1962 **Foundations of College Chemistry** Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs

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