

More aqueous equilibrium self test/practice problems, Dr. Scott Ensign.

1. **(Acid-base equilibrium review)** A solution is prepared by dissolving 0.15 moles of acetic acid in 1.00 liters of solution. K_a for acetic acid is 1.8×10^{-5} . What will be the pH of the solution once equilibrium is established?

- a. 2.0
- b. 5.6
- c. 2.8
- d. 3.9
- e. 4.7

2. **(Acid-base equilibrium review)** A solution is prepared by dissolving 0.10 moles of sodium acetate (NaCH_3COO) in 1.00 liters of solution. K_a for acetic acid is 1.8×10^{-5} . What will be the pH of the solution once equilibrium is established?

- a. 8.9
- b. 5.1
- c. 9.9
- d. 4.1
- e. 9.3

3. A solution is prepared by mixing 0.15 moles of acetic acid (CH_3COOH) with 0.10 moles of sodium acetate (NaCH_3COO) in 1.00 liters of solution. K_a for acetic acid is 1.8×10^{-5} . What will be the pH of the solution once equilibrium is established?

- a. 4.9
- b. 6.6
- c. 2.9
- d. 4.6
- e. 4.7

4. A solution is prepared by mixing 0.15 moles of acetic acid (CH_3COOH) with 0.050 moles of nitric acid (HNO_3) in 1.00 liters of solution. K_a for acetic acid is 1.8×10^{-5} . What will be the pH of the solution once equilibrium is established?

- a. 2.4
- b. 2.8
- c. 1.3
- d. 4.3
- e. 5.2

5. **(Acid-base equilibrium review)** What would be the pH of a solution prepared by dissolving 0.55 moles of NH_3 in 1.00 liters of solution? K_b for NH_3 is 1.80×10^{-5}

- a. 9.51
- b. 11.5
- c. 11.8
- d. 2.50
- e. 9.00

Aqueous Equilibrium Practice Problems

**Brian J Hathaway, Clair Murphy, Brian
Murphy**



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usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

Solvent Extraction Principles and Practice, Revised and Expanded Jan Rydberg, 2004-03-03 A complete and up to date presentation of the fundamental theoretical principles and many applications of solvent extraction this enhanced Solvent Extraction Principles and Practice Second Edition includes new coverage of the recent developments in solvent extraction processes the use of solvent extraction in analytical applications and waste re *Fundamentals of Atmospheric Modeling* Mark Z. Jacobson, 1999 Comprehensive graduate text describing the atmospheric processes numerical methods and computational techniques needed for those studying air pollution and meteorology Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **Python for Chemists** Kiyoto Aramis Tanemura, Diego Sierra-Costa, Kenneth M. Merz, Jr., 2022-08-24 Programming in Python empowers chemists to apply their domain knowledge to scales unreachable by manual effort Learning Python is easy but contextualizing chemical problems in Python is not always obvious Readers of this primer develop the skill to identify problems in their research for which code may automate operations and scale a large volume of data or calculation In addition the authors shorten the time from learning to using Python through meaningful problem sets in Chapter One **Inorganic and Organic Chemistry Multiple Choice Practice Questions (189 Pages)** E Staff, Hundreds of Inorganic and Organic Chemistry multiple choice practice questions Practice questions are divided into relevant sections for easy perusing Use this PDF to quickly assess your knowledge of Chemistry Perfect for all high school and college students and if you are preparing for standardized tests like the AP Chemistry Regents Chemistry MCAT DAT and more

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conceptual modeling The final three chapters describe the chemistry of aluminum in soils waters and watersheds These chapters illustrate the problems of spatial and temporal variability metastability and scale that continue to make aluminum geochemistry one of the great challenges in modern environmental science *General Chemistry* Donald A.

McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie s classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie s many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value

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