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Aqueous Ion Equilibrium Practice

Daniel G. Strawn, Hinrich L. Bohn, George A. O'Connor

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chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

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role Although according to its preface Volume 5 was to be the last one of the treatise these recent developments have led to a revision of that statement The present volume and its companion still in preparation deal with topics that were already mentioned in the preface to Volume 5 as gaining in importance The recent development of X ray and more particularly neutron scattering techniques have led to studies of structure in aqueous solutions of electrolytes on the one hand and to the role of water in protein structure and function on the other Both these topics have reached a stage where reviews of the present state of knowledge are useful The application of ab initio methods to calculations of hydration and conformation of small molecules has a longer history but here again a critical summary is timely The role of solvent effects in reaction kinetics and mechanisms should have had a place in Volume 2 of this treatise but as sometimes happens the author who had taken on this task failed tQ live up to his promise However since 1972 the physical chemistry of mixed aqueous solvents has made considerable strides so that the belated discussion of this topic by a new author is built on evidence that was not available at the time of publication of Volume 2

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