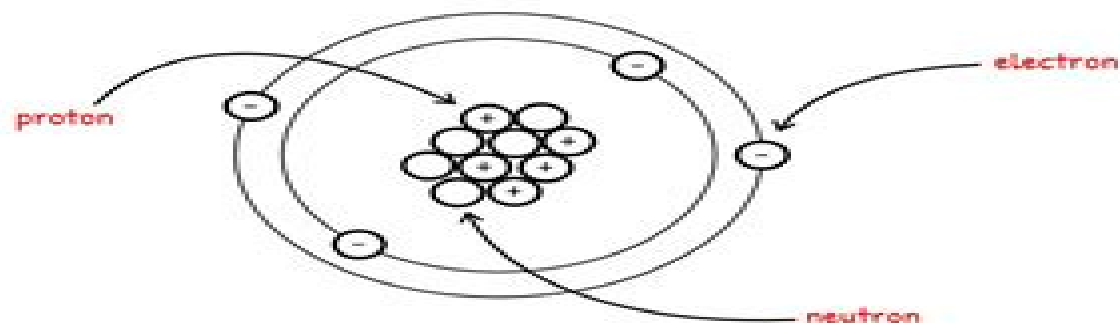


Name answer key

Date _____

Atomic Structure

Label the parts of an atom on the diagram below, and answer the questions!



1. What type of charge does a proton have?
Positive (+1)
2. What type of charge does a neutron have?
Neutral (0)
3. What type of charge does an electron have?
Negative (-1)
4. Which two subatomic particles are located in the nucleus of an atom?
Protons and neutrons
5. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
35
6. What is the atomic number of the atom in the diagram above?
5
7. What is the atomic mass/mass number of the atom in the diagram above?
(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
8. How many protons are in the nucleus of an atom with an atomic number of 15?
15 *number of protons is the same as the atomic number
9. How many electrons are in the nucleus of an atom with an atomic number of 20?
20
10. How many neutrons are in the nucleus of an atom with an atomic number of 25?
(use Periodic Table for mass) 30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30
11. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
Mass number = 6.9% (Li-Lithium has an atomic number of 3 which means it has 3 protons.
12. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
Neutrons = 11 (atomic mass: 36 - 25:atomic number)

Atomic Structure Guided Practice Problems Answer Key

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



Atomic Structure Guided Practice Problems Answer Key:

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17 Restless Earth Plate Tectonics MCQ Chapter 18 Rocks and Minerals Mixtures MCQ Chapter 19 Solar System MCQ Chapter 20 Solar System Formation MCQ Chapter 21 Space Astronomy MCQ Chapter 22 Space Science MCQ Chapter 23 Stars Galaxies and Universe MCQ Chapter 24 Tectonic Plates MCQ Chapter 25 Temperature MCQ Chapter 26 Weather and Climate MCQ The Agents of Erosion and Deposition MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Glacial deposits types angle of repose glaciers and landforms carved physical science rapid mass movement and slow mass movement The Atmosphere Composition MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Composition of atmosphere layers of atmosphere energy in atmosphere human caused pollution sources ozone hole wind and air pressure The Atmosphere Layers MCQ PDF e Book Chapter 3 practice test to solve MCQ questions on Layers of atmosphere earth layers formation human caused pollution sources and primary pollutants The Earth Atmosphere MCQ PDF e Book Chapter 4 practice test to solve MCQ questions on Layers of atmosphere energy in atmosphere atmospheric pressure and temperature air pollution and human health cleaning up air pollution global winds human caused pollution sources ozone hole physical science primary pollutants solar energy wind and air pressure and winds storms The Earth Models and Maps MCQ PDF e Book Chapter 5 practice test to solve MCQ questions on Introduction to topographic maps earth maps map projections earth surface mapping azimuthal projection direction on earth earth facts earth system science elements of elevation equal area projections equator flat earth sphere flat earth theory Geographic Information System GIS GPS latitude longitude modern mapmaking north and south pole planet earth prime meridian remote sensing science experiments science projects topographic map symbols and Venus The Earth Science and Models MCQ PDF e Book Chapter 6 practice test to solve MCQ questions on Branches of earth science geology science right models climate models astronomy facts black smokers derived quantities geoscience international system of units mathematical models measurement units meteorology metric conversion metric measurements oceanography facts optical telescope physical quantities planet earth science experiments science formulas SI systems temperature units SI units types of scientific models and unit conversion The Earthquakes MCQ PDF e Book Chapter 7 practice test to solve MCQ questions on Earthquake forecasting earthquake strength and intensity locating earthquake faults tectonic plate boundaries seismic analysis and seismic waves The Energy Resources MCQ PDF e Book Chapter 8 practice test to solve MCQ questions on Energy resources alternative resources conservation of natural resources fossil fuels sources nonrenewable resources planet earth renewable resources atom and fission chemical energy combining atoms fusion earth science facts earth s resource fossil fuels formation fossil fuels problems science for kids science projects and types of fossil fuels The Minerals and Earth Crust MCQ PDF e Book Chapter 9 practice test to solve MCQ questions on What is mineral mineral structure minerals and density minerals and hardness minerals and luster minerals and streak minerals color minerals groups mining of minerals use of minerals cleavage and fracture responsible mining rocks and minerals and science formulas The Movement of Ocean Water MCQ PDF e Book Chapter 10 practice test to solve MCQ

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Table of Contents Atomic Structure Guided Practice Problems Answer Key

1. Understanding the eBook Atomic Structure Guided Practice Problems Answer Key
 - The Rise of Digital Reading Atomic Structure Guided Practice Problems Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Guided Practice Problems Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Guided Practice Problems Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Guided Practice Problems Answer Key

- Personalized Recommendations
 - Atomic Structure Guided Practice Problems Answer Key User Reviews and Ratings
 - Atomic Structure Guided Practice Problems Answer Key and Bestseller Lists
5. Accessing Atomic Structure Guided Practice Problems Answer Key Free and Paid eBooks
- Atomic Structure Guided Practice Problems Answer Key Public Domain eBooks
 - Atomic Structure Guided Practice Problems Answer Key eBook Subscription Services
 - Atomic Structure Guided Practice Problems Answer Key Budget-Friendly Options
6. Navigating Atomic Structure Guided Practice Problems Answer Key eBook Formats
- ePub, PDF, MOBI, and More
 - Atomic Structure Guided Practice Problems Answer Key Compatibility with Devices
 - Atomic Structure Guided Practice Problems Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Atomic Structure Guided Practice Problems Answer Key
 - Highlighting and Note-Taking Atomic Structure Guided Practice Problems Answer Key
 - Interactive Elements Atomic Structure Guided Practice Problems Answer Key
8. Staying Engaged with Atomic Structure Guided Practice Problems Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Guided Practice Problems Answer Key
9. Balancing eBooks and Physical Books Atomic Structure Guided Practice Problems Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Guided Practice Problems Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Guided Practice Problems Answer Key
- Setting Reading Goals Atomic Structure Guided Practice Problems Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure Guided Practice Problems Answer Key

- Fact-Checking eBook Content of Atomic Structure Guided Practice Problems Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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