

Section A

Answers

- | | | | | | | |
|------|-------|-------|-------|-------|-------|-------|
| 1. C | 6. C | 11. A | 16. D | 21. B | 26. C | 31. B |
| 2. B | 7. D | 12. B | 17. A | 22. C | 27. B | 32. B |
| 3. B | 8. B | 13. B | 18. B | 23. B | 28. C | 33. B |
| 4. A | 9. D | 14. C | 19. A | 24. D | 29. B | |
| 5. D | 10. A | 15. B | 20. C | 25. D | 30. A | |

Solution

- C

Absolute value of the slope of the graph = $\frac{P}{mc} \propto \frac{1}{mc}$ (as the power is the same for both liquid)

 - ✓ (1) As the absolute value of the slope of X is smaller, the heat capacity mc is greater.
 - ✗ (2) Although the value of mc for X is greater, but the mass m may not be the same, thus the relation of specific heat capacity c cannot be determined.
 - ✓ (3) If both of them are water, then their specific heat capacity c must be the same, thus the mass m of X is greater.
- B

 - ✗ (1) Mist formation is the condensation of vapour into liquid, thus latent heat of vaporization should be released to the glasses.
 - ✓ (2) The molecular potential energy liquid is lower than that of vapour.
 - ✗ (3) Mist formation can occur at temperatures even below the room temperature.
- B

 - ✓ (1) As the pressure is constant, volume of gas increases as the temperature increases. Thus the gas molecules become more separated and the density decrease.
 - ✓ (2) As the temperature increases, kinetic energy and average speed of gas molecules increase. Thus, the momentum of each gas molecule increases. Each gas molecule then hit the wall with a greater force due to its greater momentum.
 - ✗ (3) The gas molecules hit the walls of the cylinder more violently as temperature increases, thus the molecules must hit the walls of the cylinder less frequently to give the same pressure.
- A

In 1 m^3 , the number of mole of gas molecules n is found by : $P V = n R T$
 $\therefore (250 \times 10^3) (1) = n (8.31) (25 + 273) \quad \therefore n = 100.95$
 Number of gas molecules in $1 \text{ m}^3 = 100.95 \times 6.02 \times 10^{23} = 6.1 \times 10^{25}$

14 Dse Phy Answer

J Ma



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Table of Contents 14 Dse Phy Answer

1. Understanding the eBook 14 Dse Phy Answer
 - The Rise of Digital Reading 14 Dse Phy Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying 14 Dse Phy Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 14 Dse Phy Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from 14 Dse Phy Answer
 - Personalized Recommendations
 - 14 Dse Phy Answer User Reviews and Ratings
 - 14 Dse Phy Answer and Bestseller Lists
5. Accessing 14 Dse Phy Answer Free and Paid eBooks
 - 14 Dse Phy Answer Public Domain eBooks
 - 14 Dse Phy Answer eBook Subscription Services
 - 14 Dse Phy Answer Budget-Friendly Options

6. Navigating 14 Dse Phy Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - 14 Dse Phy Answer Compatibility with Devices
 - 14 Dse Phy Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 14 Dse Phy Answer
 - Highlighting and Note-Taking 14 Dse Phy Answer
 - Interactive Elements 14 Dse Phy Answer
8. Staying Engaged with 14 Dse Phy Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 14 Dse Phy Answer
9. Balancing eBooks and Physical Books 14 Dse Phy Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 14 Dse Phy Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 14 Dse Phy Answer
 - Setting Reading Goals 14 Dse Phy Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 14 Dse Phy Answer
 - Fact-Checking eBook Content of 14 Dse Phy Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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