# Strengths of Acids and Bases

- Strengths of Acids
  - Strong acid acid that ionizes completely
    - ☐ Produce the maximum number of ions
    - □ Good conductors of electricity
    - □ Ex: hydrochloric acid (HCl)
  - Weak acid acid that ionizes only partially
    - □ Some of the acid remains in molecular form
    - □ Ex: acetic acid (vinegar)
  - Using the Bronsted-Lowry model, if the reactant acid is strong, its conjugate base is weak

## 19 3 Chemistry Answers Strengths Of Acids And Bases

Neil D. Jespersen, Pamela Kerrigan

#### 19 3 Chemistry Answers Strengths Of Acids And Bases:

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