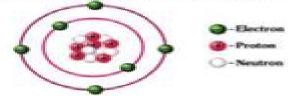
EQ: Describe atomic structure.

- Atoms are made of extremely tiny particles called protons, neutrons, and electrons.
- Protons and neutrons are in the center of the atom, making up the nucleus.
- Electrons surround the nucleus.
- Protons have a positive charge.
- Electrons have a negative charge.
- The charge on the proton and electron are exactly the same size but opposite.
- Neutrons have no charge.
- Since opposite charges attract protons and electrons attract each other.
 - Label the nucleus (protons, neutrons) and electrons in the drawing of a carbon atom.
 - Draw a line between the subatomic particle and its charge.

proton	
electron	
neutron.	

no charge positive charge negative charge

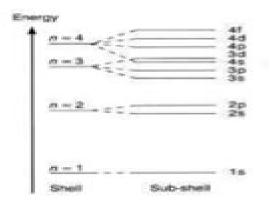


- The electrons surrounding an atom are located in regions around the nucleus called "energy levels".
- An energy level represents the 3-dimensional space surrounding the nucleus where electrons are most likely to be.
- The first energy level is closest to the nucleus. The second energy level is a little farther away than the first.
 The third is a little farther away than the second, and so on.
- Each energy level can accommodate or "hold" a different number of electrons before additional electrons begin to go into the next level.
- When the first energy level has 2 electrons, the next electrons go into the second energy level until the second level has 8 electrons.
- When the second energy level has 8 electrons, the next electrons go into the third energy level until the third level has 8 electrons.
- When the third energy level has 8 electrons, the next 2 electrons go into the fourth energy level.
- The electrons in the energy level farthest from the nucleus are called valence electrons.
- Atoms in the same column (group) in the periodic table have the same number of valence electrons.

Energy levels

Electrons surround the nucleus of an atom in regions energy levels. Even though atoms are spherical, the levels in an atom are more easily shown in concentric

Energy Level	Maximum # electrons
1	2
2	8
3	18
4	32
5	50
6	72



called energy circles.

Atomic Structure Study Guide

Albert A Gayle

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