

ATOMIC STRUCTURE QUIZ



Multiple Choice: Read the question carefully and select the choice that best answers the question. 2 pt each

1.) The charge of a proton is...

- a.) Negative
- b.) Positive
- c.) Neutral
- d.) Unknown

2.) What does the atomic number tell you?

- a.) Number of neutrons
- b.) The size of the atom
- c.) The mass of the atom
- d.) The number of protons

3.) How can you find the mass of an isotope?

- a.) Adding protons and neutrons
- b.) Adding protons and electrons
- c.) Subtracting protons and neutrons
- d.) Subtracting electrons and neutrons

4.) What is the Atomic Mass of Lithium?

- a.) 6.94
- b.) 3
- c.) 4
- d.) 6.026

5.) The charge of an electron is...

- a.) Negative
- b.) Positive
- c.) Neutral
- d.) Unknown

6.) In a Bohr model, how many electrons can go in the first orbital?

- a.) 8
- b.) 16
- c.) 2
- d.) 3

7.) What is the element symbol for Radon?

- a.) Rb
- b.) Ra
- c.) Rn
- d.) Ru

8.) Which subatomic particle does not have a mass?

- a.) Neutron
- b.) Proton
- c.) Neutron
- d.) Electron

9.) In a neutral atom, the number electrons equals the number of...

- a.) Neutrons
- b.) Protons
- c.) Mass
- d.) Cannot Determine

10.) What is the name of the element with the symbol, Hg

- a.) Mercury
- b.) Bromine
- c.) Iodine
- d.) Helium

Atomic Structure Test Questions

GBP Editorial



Atomic Structure Test Questions:

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In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Atomic Structure Test Questions** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of people who partake in its reading experience.

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