

**Atoms and Bonding • Reading/Taking Guide****Ionic Bonds** (pp. 184–189)

*This section explains how atoms become ions. It describes the electrical charge of an ionic compound and what the chemical formula of an ionic compound tells you.*

**Use Target Reading Skills**

*As you read, fill in the table to compare and contrast the properties of sodium chloride with the properties of its component ions.*

|                             | Color                | State at Room Temperature | Stability                     |
|-----------------------------|----------------------|---------------------------|-------------------------------|
| Sodium chloride             |                      |                           |                               |
| Elements of sodium chloride | sodium:<br>chlorine: | sodium:<br>chlorine:      | sodium: unstable<br>chlorine: |

**Ions** (pp. 185–186)

1. An atom or group of atoms that has an electric charge is called a(n) \_\_\_\_\_.

2. What happens to an atom when it loses an electron?

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3. What happens to an atom when it gains an electron?

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# Atoms And Bonding Notetaking Guide Answers

**Rachel S Tattersall**



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