## Atomic structure linkword

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#### CLUES

9 down	An atom is composed of equal numbers of and electrons. (7)						
10 across	An ion has a different number of protons and, and so it is charged. (9)						
8 down	Proton number is also called " number" and has the symbol Z. (6)						
7 across	number is also called "nucleon number" and has the symbol A. (4)						
3 down	The number of neutrons in each isotope is given by $A = \dots = Z$ . (5)						
6 down	Two versions of the same element will have different masses and different numbers of but the same number of protons. (8)						
11 across	Each version of an element with the same proton number but different mass numbers is known as an (7)						
4 across	Electrons occupy the space around the (7)						
5 down	The nucleus lies at the of the atom and contains only protons and neutrons. (6)						
2 down	Protons and neutrons collectively are sometimes called (8)						
1 dicean	The nucleus is time positive and (5)						

WORD BANK: PROTONS, NUCLEUS, NUCLEONS, NEUTRONS, MINUS, MASS, ISOTOPE, ELECTRONS, DENSE, center, ATOMIC

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