Study Guide Atomic Structure and Periodic Table

Circle the possible best answer for each question: (1pt each)

Neutrons are subatomic particles that are located

- a. In the nucleus and have a positive charge
- b. In the nucleus and have a negative charge
- c. In the nucleus and have no charge
- d. Around the nucleus and have a positive charge

The atomic number of an element is equal to the number of

- a. Electrons
- b. Neutrons
- c. Protons
- d. Protons plus electrons

Which of the pair below in equal to each other in an atom

- a. Protons and electrons
- b. Neutrons and protons
- c. Neutrons and electrons
- d. Mass number and electrons

In the late 1800s, experiments using cathode ray tubes led to the discovery of the

- a. Positron
- b. Neutron
- c. Proton
- d. Electron

Which to particles each have a mass approximately equal to one atomic mass unit?

- a. Electrons and positron
- b. Proton and electron
- c. Electron and neutron
- d. Proton and neutron

To which group of the periodic tables do lithium and potassium belong?

a) Alkali metals b) Alkaline earth metals c) transition metals d) halogens

To which group of the periodic tables do Fluorine and Chlorine belong?

a) Alkali metals
 b) Alkaline Earth metals
 c) Transition metals
 d) halogens

Atomic Structure And Periodic Relationships Study Guide

Barron's Educational Series, Mark Kernion, Joseph A. Mascetta

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