1. Define the terms acid and base using the Arrhenius and Bronsted/Lowry definitions.

| Arrhenius | | Bronsted/Lowry | |
|-----------|--|----------------|--|
| Acid | | | |
| Base | | | |

Acids and bases share some properties but are different in others. Using your knowledge of acids and bases, fill in the column in the table below.

| Property | Acids | Bases |
|-------------------------------------|-------|-------|
| Conductivity in Aqueous Solution | | |
| Taste | | |
| Feel | | |
| Reaction with Metal | | |
| Reaction with Carbonate | | |
| Color with Litmus Paper | | |
| Color with Universal Indicator | | |
| Color with Phenolphthalein | | |

- Using your knowledge of the Brønsted-Lowry theory of acids and bases, complete the equations for the following acid-base reactions and indicate each conjugate acid-base pair.
 - a) HNO, + OH ----
 - b) CH₃NH₂ + H₂O →

Acid And Bases Chemistry Workbook Answers

Daniel F McAuley

Acid And Bases Chemistry Workbook Answers:

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