

Titrant

	Electrolyte	Concentration, M
X	NaOH	0,1

Maximal titrant volume (ml): 19

Examples

Strong acid and strong base	1	2	3
Weak acid and strong base	4	5	6
Weak diprotic acid and strong base	7	8	9
Weak base and strong acid	10	11	12
Weak base (salt) and strong acid	13	14	15
Mixture of acids and strong base	16	17	18



Calculate



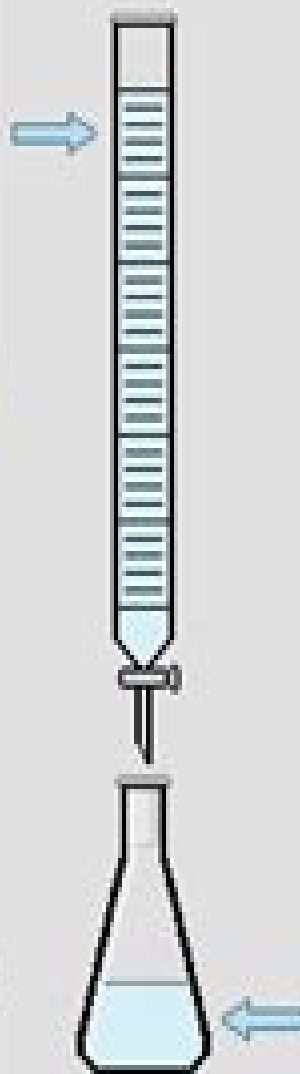
Clear all



Exit



About



pH = 2,88

Solution to be titrated

	Electrolyte	Concentration, M
X	CH ₃ COOH	0,1
X		
X		
X		
X		
X		
X		
X		

Solution volume (ml): 13

Indicators

X	Phenol-phthalein
X	

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**National Institutes of Health (U.S.).
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