

CHAPTER 26 : ACID-BASE

Total body water volume: _____ this is 60% of body weight

Water: universal solvent

Solutes: nonelectrolytes and electrolytes

Extracellular and intracellular

- ECF : the fluid on the _____ the cell major cation is sodium and major anion is chloride
- Sodium is the most abundant cation in the extracellular fluid
- ICF: fluid _____ the cell major cation is potassium

Water intake is _____ to water output 2500 ml/day

Water output: urine, insensible water loss, perspiration, and feces

Water reabsorption: in collecting ducts proportional to ADH release

Aldosterone: important in regulation of sodium ion concentration

*water follows sodium

Decrease ADH: diluted urine and decrease in volume in body fluids

Increase ADH: concentrated urine

Atrial natriuretic peptide: decrease BP and blood volume

Important buffer systems

- Respiratory system
- Kidneys

normal pH of body fluids

- Arterial blood = pH 7.4
- Venous blood and IF fluid = pH 7.35
- ICF = pH 7.0
- Alkalosis or alkalemia = arterial blood pH > 7.45
- Acidosis or acidemia = arterial pH < 7.35
- Physiological acidosis = pH between 7.0 - 7.35

Hydrogen ion regulation

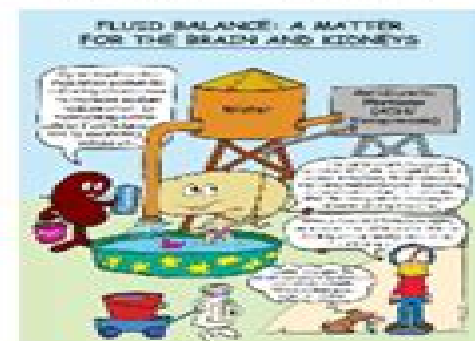
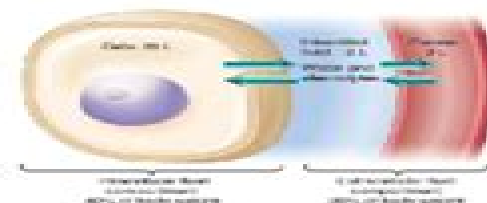
1. Chemical buffer system _____ (rapid) first line of defense
2. Respiratory centers - in brain stem act within 1 - 3 min
3. Renal mechanisms - _____ most potent

Three major chemical buffers

Chemical buffer: resist change in pH changes when a strong acid or base is added

1. Bicarbonate buffer system
2. Phosphate buffer system
3. Protein buffer system

Major fluid compartments in the body



Acid And Base Study Guide Answers

David Canon



Acid And Base Study Guide Answers:

Critical Care Study Guide Gerard J. Criner, Gilbert E. D'Alonzo, 2013-03-09 Critical care medicine is a dynamic and exciting arena where complex pathophysiologic states requiring extensive knowledge and clinical acumen are commonly found. Caring for critically ill patients requires an extensive knowledge of basic pathophysiology as well as awareness of the appropriate diagnostic tests and therapeutic interventions. Because this knowledge base crosses many different disciplines, introduction to caring for the intensive care patient while exciting may also be intimidating. This textbook is designed toward making the dynamic environment of the critical care unit understandable and the approach to the patient both logical and successful. The book contains three components: 1. description of the procedural tasks commonly performed for the critically ill patient; 2. explanation of the most common pathophysiologic states encountered; and 3. description of specific disease entities with details of their differential diagnosis, diagnostic strategy, and therapeutic plan. This book also uses several educational approaches that we have found useful in our own teaching sessions and prior textbooks. Clinical cases introduce chapters and highlight chapter segments to emphasize clinical relevancy. As an additional study aid, margin notes highlight important teaching points and facilitate easy review of chapter content. To consolidate the principles outlined in each chapter, review questions with full text explanations are provided at the end of each chapter. All these elements help reinforce the most important messages for the reader. *Study Guide with Selected Solutions* Karen Timberlake, 2002-08-06 Written by the author, the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams. Also contains complete solutions to odd-numbered problems.

Organic Chemistry, Study Guide and Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, 2000 On the cover of this book is a Pacific yew tree found in the ancient forests of the Pacific Northwest. The bark of the Pacific yew tree produces Taxol, found to be a highly effective drug against ovarian and breast cancer. Taxol blocks mitosis during eukaryotic cell division. The supply of Taxol from the Pacific yew tree is vanishingly small; however, a single 100-year-old tree provides only about one dose of the drug, roughly 300 mg. For this reason, as well as the spectacular molecular architecture of Taxol, synthetic organic chemists fiercely undertook efforts to synthesize it. Five total syntheses of Taxol have thus far been reported. Now a combination of isolation of a related metabolite from European yew needles and synthesis of Taxol from that intermediate supply the clinical demand. This case clearly demonstrates the importance of synthesis and the use of organic chemistry. It is just one of the many examples used in the text that will spark the interest of students and get them involved in the study of organic chemistry. *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein, Laurie S. Starkey, 2025-03-04 Success in organic chemistry requires mastery in two core aspects: fundamental concepts and the skills needed to apply those concepts and solve problems. With *Organic Chemistry Student Study Guide and Solutions Manual* 5th Edition, students can learn to become proficient at approaching new situations methodically based on a repertoire of skills.

These skills are vital for successful problem solving in organic chemistry

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promote locally sourced foods in their cafeteria and school meals

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a lot of sense for schools it provides great publicity and attracts a higher level of student it exemplifies the lessons on
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jul 14 2016 even children 0 2 years old can be positioned to observe a chef at work and enjoy banging a wooden spoon on a
pot to feel a part of the process children ages 2 3 years old can become involved by washing fruits and vegetables mixing
batters and wiping up the counter children ages 4 5 years old can begin cracking open an egg using the mixer

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teaching kids where their food comes from is an important step in developing an understanding of healthy eating it s also an
important lesson about agriculture industry and how farming sustains the economy many children never visit a farm and
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