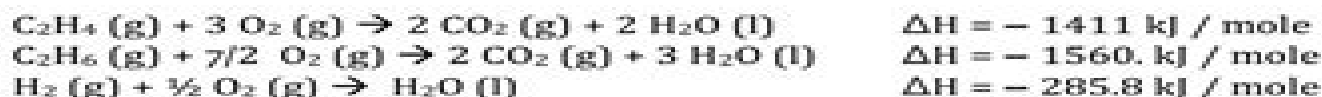


CHEM 12 Hess's Law Worksheet

1. Calculate the ΔH for the reaction: $\text{C}_2\text{H}_4 (\text{g}) + \text{H}_2 (\text{g}) \rightarrow \text{C}_2\text{H}_6 (\text{g})$,
from the following data:



2. Calculate the ΔH for the reaction:



3. Find the ΔH_f° (heat of formation) for acetic acid, $\text{HC}_2\text{H}_3\text{O}_2$, using the following thermochemical data:



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**Richard M. Felder, Ronald W.
Rousseau, Lisa G. Bullard**



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