ND SEMESTER EXAMINATION 2021-2022

CHEMISTRY PAPER 1 (THEORY)

M.M:70

(Time allowed : Three hours)

| 2 | | | | | | | |
|-------|------------------------------------------------------------------|---------------------------------------------------------------------|-----------------------|--------|-------------------------|--------------------|--|
| Gene | ral Instru | ctions : | | | | | |
| 1. / | All questions are compulsory | | | | | | |
| 2. 7 | The intended marks for questions are given in brackets | | | | | | |
| 5. E | salanced | equations mus | t be given wh | ereve | r possible and diagra | ms | |
| * | where the | y are neipiul. | | | | | |
| 4. \ | When sol | ving numerical | problems ,all | esse | ntial working must b | e shown. | |
| Q1.(a |) Fill in t | he blanks: | | | | [4×1] | |
| (i) | The conjugate acid of HPO32- is | | | | | | |
| (ii) | Precipitation of a compound take place when ionic product | | | | | | |
| | 1S | than its so | plubility produ | ct. | | | |
| (iii) | Alkaline earth metals arereducing agents than alkali metale | | | | | | |
| (iv) | Nucleophiles are electronspecies. | | | | | | |
| 71- |) Compl | ete the follows: | | | | | |
| 100 | from the | e choices give | ng statements i | by sei | lecting the correct alt | | |
| (i) | from the choices given: A racemic compound is- | | | | | [4×1] | |
| (a) | Dextro- | rotatory | 4.2 | (b) | I sava vatata | | |
| (c) | | y inactive mix | ture | (d) | | CARCINO CONTRACTOR | |
| | | | | (4) | Optically active mix | ture | |
| (ii) | Among the following a lewis acid is- | | | | | | |
| (a) | PCl ₃ | (b) NCl ₃ | (c) AlCl ₃ | | d) AsCl ₃ | | |
| | | | | | | | |
| (iii) | (a) Crystallization of sucrose from solution (b) Rusting of iron | | | | | | |
| (a) | | | | | | | |
| (0) | Melting of ice (d) Vaporization of cample | | | | | amphor | |
| (iv) | The pre- | The pressure at which one male of a second | | | | | |
| | litre is- | The pressure at which one mole of gas at 0°C occupies a volume of a | | | | | |
| | | | | | | | |

(c) 11.2 atm (d) 22.4 atm

(a) 2.24 atm (b) 4.48 atm

Nd Semester Exam Chemistry Form

Michael Brown

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