

Exam #10 Review: Acids, Bases, and pH

1. Know the properties of acids and bases. What are their similarities and differences?

Acids

- Taste *sour*
- React with metals to form hydrogen gas
- Contain a hydrogen ion (H^{+1})
- pH value less than 7

Bases

- Taste *bitter*
- Feel slippery
- Contain a hydroxide ion (OH^{-1})
- pH value greater than 7

BOTH

- change colors of indicators
- react with each other to form salt and water
- conduct electricity when dissolved in solution (electrolytes)

2. Know the rules for naming
- acids**
- (binary acid and oxyacid) and name the following below:

Binary acids (only 2 elements): hydro – element – ic acidOxyacids (contain a polyatomic ion): NO "HYDRO" IN THE NAME

- -ate ion $\rightarrow$ -ic acid
- -ite ion $\rightarrow$ -ous acid

REMEMBER:

"I ate something icky,
all nite I was nauseous"

- a. HCl **hydrochloric acid**
b. HF **hydrofluoric acid**
c. $HClO_2$ **chlorous acid**

- d. HNO_3 **nitric acid**
e. H_3PO_4 **phosphoric acid**
f. H_2SO_3 **sulfurous acid**

3. Know the rules for naming
- bases**
- and name the following below:

Element + hydroxide

- a. NaOH **sodium hydroxide**
b. LiOH **lithium hydroxide**

- c. $Al(OH)_3$ **aluminum hydroxide**
d. $Ba(OH)_2$ **barium hydroxide**

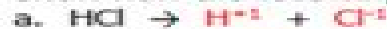
4. How are Arrhenius acids and bases defined?

Acids – produce H^{+1} ions in aqueous solutionsBases – produce OH^{-1} ions in aqueous solutions

5. How are Bronsted-Lowry acids and bases defined?

Acids – donate (give away) a H^{+1} ion (proton)Bases – accept (take in) a H^{+1} ion (proton)

6. Show how the following acids and bases ionize:



7. If it is a strong acid or base, how much do they ionize in solution?

STRONG acids & bases COMPLETELY ionize (all break apart) in a solution

8. If it is a weak acid or base, how much do they ionize in solution?

WEAK acids & bases do NOT completely ionize

9. What is a neutralization reaction? What products are always produced?

Acid reacts with a basealways produces water and a salt

10. Complete the following neutralization reaction and what type of salt is made:

Type of Salt: neutralType of Salt: neutral

Acids And Bases Exam Review Guide Answers

Arshad Iqbal



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Examination Linda Anne Silvestri, 2011 Accompanying CD ROM includes more than 4500 NCLEX examination style review questions in quiz study or exam mode a diagnostic pre test audio and video questions and audio review summaries

Elsevier's Canadian Comprehensive Review for the NCLEX-RN® Examination - E-Book Patricia A. Bradley, Karin L. Page-Cuttrara, Linda Anne Silvestri, Angela Silvestri, 2023-03-28 A truly Canadian edition of Elsevier's best selling NCLEX exam review book Elsevier's Canadian Comprehensive Review for the NCLEX RN Examination 3rd Edition provides everything you need to prepare for the NCLEX exam complete content review more than 5 000 NCLEX practice questions in the book and online and preparation for the Next Generation NCLEX In addition all answers include detailed rationales and test taking strategies with tips on how to best approach each question Integrating Canadian approaches to nursing throughout the text this book is the only comprehensive NCLEX review written from a Canadian perspective It's THE book of choice for NCLEX preparation Completely up to date coverage from a Canadian perspective reflects Canadian approaches to nursing and health care including the addition of the latest Canadian statistics research legislation regulations references clinical practice guidelines and more More than 5 000 practice questions in the text and online offer ample testing practice UNIQUE Detailed test taking strategy and rationale is included for each question offering clues for analyzing and uncovering the correct answer option UNIQUE Priority Nursing Action boxes provide information about the steps to be taken in clinical situations requiring clinical judgement and prioritization UNIQUE Pyramid Points icons indicate important information identifying content that typically appears on the NCLEX RN examination UNIQUE Pyramid Alerts appear in red text and highlight important nursing concepts New graduate's perspective is offered on how to prepare for the NCLEX RN in addition to nonacademic preparation the CAT format and test taking strategies Mnemonics are included to help you remember important information 79 question comprehensive exam covers all content areas in the book in the same percentages that they are covered on the actual NCLEX RN test plan and includes four case study format questions for the NGN Practice questions on delegation prioritization and triage disaster management emphasize these areas on the NCLEX exam

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