

WORKSHEET

Acid-base Titration I

Maximum score: 100 points

Your Name:

Partner:

Lab Section (Day & Time):

1. Attach your annotated plots and label the equivalence point and the half-equivalence points appropriately for the titration curves for HCl, KHP, and one trace for the unknown acid, and record the data from the plots. (30 points)

Acid Trial	pH at start of titration	pK _a or pH at half-equivalence point	pH at equivalence point	pH at end of plot
HCl	2.2	XXXXXXXXXXXX	5.3	11.75
KHP	4.6	5.1	10.5	11.70
Unknown	2.3	2.9	9.0	11.65

2. Complete the table using data from your lab notebook. Calculate the acid concentration using $M_a V_a = M_b V_b$. The unknown acid will require a correction for the dilution. (20 pts).

Acid Trial	Volume acid (mL)	NaOH Concentration (M)	Volume of base at Equivalence point (mL NaOH)	Known Acid Concentration (M)	Calculated Acid Concentration (M)
HCl	55	.1111	4.43	.1133	.0089
KHP	55	.1111	2.54	.0558	.0051
Diluted Unknowns Trial 1	55	.1111	9.005	XXXXXXXXXX	.0081
Diluted Unknowns Trial 2				XXXXXXXXXX	

Will not do second trial for unknown

Acid Base Titration Worksheet Chemistry If8766

Clifford W. Hand, Harry Lyon Blewitt



Acid Base Titration Worksheet Chemistry If8766:

Acid-base Titrations in Nonaqueous Solvents James Sherwood Fritz, 1973 *Aqueous Acid-base Equilibria and Titrations* Robert De Levie, 1999 This book will give students a thorough grounding in pH and associated equilibria material absolutely fundamental to the understanding of many aspects of chemistry It is in addition a fresh and modern approach to a topic all too often taught in an out moded way This book uses new theoretical developments which have led to more generalized approaches to equilibrium problems these approaches are often simpler than the approximations which they replace Acid base problems are readily addressed in terms of the proton condition a convenient amalgam of the mass and charge constraints of the chemical system considered The graphical approach of Bjerrum Hagg and Sillen is used to illustrate the orders of magnitude of the concentrations of the various species involved in chemical equilibria Based on these concentrations the proton condition can usually be simplified often leading directly to the value of the pH In the description of acid base titrations a general master equation is developed It provides a continuous and complete description of the entire titration curve which can then be used for computer based comparison with experimental data Graphical estimates of the steepness of titration curves are also developed from which the practicality of a given titration can be anticipated Activity effects are described in detail including their effect on titration curves The discussion emphasizes the distinction between equilibrium constants and electrometric pH measurements which are subject to activity corrections and balance equations and spectroscopic pH measurements which are not Finally an entire chapter is devoted to what the pH meter measures and to the experimental and theoretical uncertainties involved

Volumetric Analysis Ralf Martens-Menzel, Lena Harwardt, Hanns-Jürgen Krauss, 2025-05-19 This standard work on volumetric analysis based on the 20th German edition provides comprehensive information on the theory of acid base titration redox titration complexation titration and precipitation titration with both classical and instrumental indication of the equivalence point Many applications are described and explained in detail with examples in pharmaceutical and environmental analysis

Acid-bases in Analytical Chemistry Izaak Maurits Kolthoff, 1966 **Titration in Non-aqueous Media** István Gyenes, 1968 **Acid-Base Diagrams**

Heike Kahlert, Fritz Scholz, 2013-07-31 Understanding acid base equilibria made easy for students in chemistry biochemistry biology environmental and earth sciences Solving chemical problems be it in education or in real life often requires the understanding of the acid base equilibria behind them Based on many years of teaching experience Heike Kahlert and Fritz Scholz present a powerful tool to meet such challenges They provide a simple guide to the fundamentals and applications of acid base diagrams avoiding complex mathematics This textbook is richly illustrated and has full color throughout It offers learning features such as boxed results and a collection of formulae

Ph Scale And Acid-base Titrations/standardization Postma Roberts Hollenberg, James M. Postma, Julian Roberts, J. Leland Hollenberg, 2000-01-15

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Reactions of Acids and Bases in Analytical Chemistry Adam Hulanicki, 1987 **Lecture Demonstrations Employing Mechanical Reproductions of Acid-base Titration Curves and Photometric Titration Curves** Sondra L. Spannuth, 1974 **Acid-base Chemistry** Clifford W. Hand, Harry Lyon Blewitt, 1986 Very Good No Highlights or Markup all pages are intact **Acid Base Titrations** Shah Purvesh, 2015-08-04 In Acid Base Titrations the author discussed various acid base titration It gives some basic concept of acid base titration The analysis of acid base titration discussed in this book It also covers Titration of polyprotic acid and mixture of acids Titration of a Polybasic base with a Strong Acid pH transition range for an indicator titration of salts Differential Alkali titration are discussed in simplest type **Acid-Base Titration in Non-Aqueous Solvents** James S. Fritz, 1973-06-01 *General Theory of Acid-base Titration* Pat Richard Pondy, 1949

Identification of Misconceptions Concerning Acid-base Titrations Lockwood Ryerson Gibbs, 1995 **EDTA Titrations** H. A. Flaschka, 2013-10-22 **EDTA Titrations An Introduction to Theory and Practice** Second Edition considers the theoretical background full procedural details and some practical applications of EDTA titrations Ethylenediamine tetra acetic acid EDTA has risen from an obscure chemical compound to the most widely used organic reagent This book is composed of 21 chapters The opening chapters present the general theoretical foundations of EDTA titrations The subsequent chapters describe the properties of EDTA such as the stability constants titration curves selectivity and masking effect These topics are followed by discussions on titration types standard solutions and reagents The remaining chapters cover some of the practical applications of EDTA titrations This book is directed toward students with advanced courses in analytical and organic chemistry **Monitoring Acid-Base Titrations With a Ph Meter** John W. Alcock, Marcia L. Gillette, 1997-02-01

Working with Chem Separate Wink, 1999-12-01 **PH, Buffers and Acid-base Titrations** LaMotte Chemical

Products Co. Educational Products Division, 197?
Johansson, 1978

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