FINAL EXAM Summary Sheet

Potential Study Questions

<u>Disclaimer</u>: The following are potential topics for exam questions and sample questions; the exam may or may not contain these questions, and questions not included in this sheet may or may not be on the exam. There also may be typos in this sheet.

ALKANES

Physical Properties

- Hydrocarbons, dispersion forces, non-polar, not miscible w/water, low density
- sp³ hybridized carbon: C-C, C-H bonds, electronegativity, MO theory & orbitals
- m.p.: odd/even effect.
- b.p. C1-4: gases, C5-17: liquids, C18 and up: solids
- linear, branched, cyclics, chirality (incl. optical activity, D/L, R/S, d/l,+/-),
- non-equivalent carbons
- isomers: constitutional, conformational isomers, stereo-isomers
- naming, special groups:
 - o Me, Et, Pr, Bu
 - o iPr, iBu, sec-Bu, tBu
 - o neopentyl

<u>Task</u>: Name compounds 1 and 2; identify any chiral centers and # of chemically non-equivalent carbons; where would the compounds absorb in IR in UVvis? Identify the participating orbitals.



- Rotational/conformational isomers = conformations
 - o eclipsed
 - values: H/H, CH3/H, CH3/CH3
 - o staggered, anti, gauche
 - value: gauche: CH3/CH3

Task: draw an energy rotational energy plot (360°) for pentane along C2-C3 bond

- Cycloalkanes:
 - o Spiro, fused, bridged: naming
 - o Heats of combustion: ring strains (angle, torsional, sterics)
 - o Cis/trans
 - 1,3-diaxial interactions
 - value H/CH3
 - Ring-flip for cyclohexane

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G Psacharopoulos

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in sacramento ca ramos studied art at sacramento state college where he met and studied under wayne thiebaud american b 1920 who would become a lifelong friend

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