

Strengths of Acids and Bases

☐ Strengths of Acids

☐ Strong acid – acid that ionizes completely

- ☐ Produce the maximum number of ions
- ☐ Good conductors of electricity
- ☐ Ex: hydrochloric acid (HCl)

☐ Weak acid – acid that ionizes only partially

- ☐ Some of the acid remains in molecular form
- ☐ Ex: acetic acid (vinegar)

☐ Using the Bronsted-Lowry model, if the reactant acid is strong, its conjugate base is weak

19 3 Chemistry Answers Strengths Of Acids And Bases

Peter Atkins



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