

# ASSIGNING OXIDATION NUMBERS – WORKSHEET 1

## Part A:

In the following questions, give the oxidation number of the indicated atoms/ion.

1. N in  $\text{N}_2\text{O}_3$  \_\_\_\_\_

2. S in  $\text{H}_2\text{SO}_4$  \_\_\_\_\_

3. C \_\_\_\_\_

4. C in CO \_\_\_\_\_

5. Na in NaCl \_\_\_\_\_

6. H in  $\text{H}_2\text{O}$  \_\_\_\_\_

7. Ba in  $\text{BaCl}_2$  \_\_\_\_\_

8. N in  $\text{NO}_2^-$  \_\_\_\_\_

9. S in  $\text{Al}_2\text{S}_3$  \_\_\_\_\_

10. S in  $\text{HSO}_4^-$  \_\_\_\_\_

11. Cl in  $\text{Fe}(\text{ClO}_2)_3$  \_\_\_\_\_

12. Fe in  $\text{Fe}(\text{ClO}_2)_3$  \_\_\_\_\_

13. N in  $\text{NO}_3^-$  \_\_\_\_\_

14.  $\text{Cu}^{2+}$  \_\_\_\_\_

15.  $\text{Zn}^{2+}$  \_\_\_\_\_

16. C in  $\text{CH}_4$  \_\_\_\_\_

17. Mn in  $\text{MnO}_2$  \_\_\_\_\_

18. S in  $\text{SO}_3^{2-}$  \_\_\_\_\_

19.  $\text{Mg}^{2+}$  \_\_\_\_\_

20.  $\text{Cl}^-$  \_\_\_\_\_

21.  $\text{O}_2$  \_\_\_\_\_

22.  $\text{P}_4$  \_\_\_\_\_

23. Na in  $\text{Na}_2\text{S}$  \_\_\_\_\_

24. S in  $\text{H}_2\text{S}$  \_\_\_\_\_

25.  $\text{Ca}^{2+}$  \_\_\_\_\_

26. C in  $\text{CN}^-$  \_\_\_\_\_

27. H in  $\text{OH}^-$  \_\_\_\_\_

28. Mn in  $\text{KMnO}_4$  \_\_\_\_\_

29. I in  $\text{Mg}(\text{IO}_3)_2$  \_\_\_\_\_

30. C in  $\text{C}_2\text{O}_4^{2-}$  \_\_\_\_\_

# Assigning Oxidation Numbers Answers

**J Spring**



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