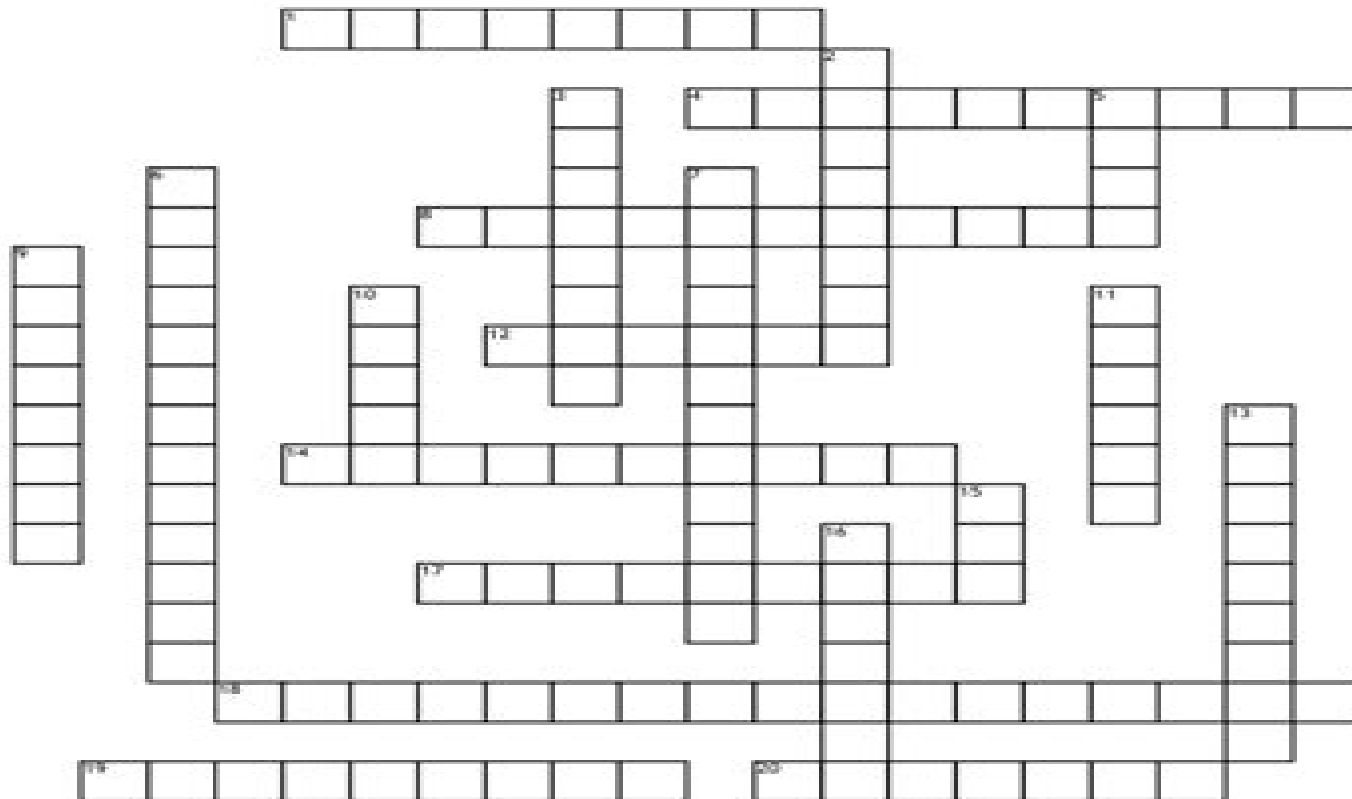


Atomic Theory



Across

1. Negative charge
4. Discovered it is impossible to know exact position and momentum of a particle in quantum physics
8. Discovered that electrons orbit or not always circular
12. Positive charge
14. Develops idea of atom
17. Discovered elements
18. Two atoms with the same element chemically bonded together

19. Said they were four elements

20. Neutral charge

Down

2. Discovered the charge and mass of electrons
3. Developed nuclear atom model
5. Found that electrons orbit
6. A group of elements with an overall charge
7. An atom with an overall charge consisting of one element
9. Discovered neutrons

10. Defined element as any substance that can't be decomposed to a simpler substance

11. Came up with the four components of an atom

13. Credited for creating the first version of the modern periodic table

15. A charged atom

16. An atom with a different number of neutrons

Atomic Theory Crossword

S. K. Gupta & Anubhuti Gangal

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the authors' names.

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years before Computer Science was added to the department's name such pioneering results in computation and control as Vannevar Bush's Differential Analyzer early cybernetic devices and numerically controlled servomechanisms the Whirlwind computer and the evolution of time sharing computation had already been achieved *Chemistry for High School*,

Atomic Physics C.J. Foot, 2005 This text will thoroughly update the existing literature on atomic physics Intended to accompany an advanced undergraduate course in atomic physics the book will lead the students up to the latest advances and the applications to Bose Einstein Condensation of atoms matter wave interferometry and quantum computing with trapped ions The elementary atomic physics covered in the early chapters should be accessible to undergraduates when they are first introduced to the subject To complement the usual quantum mechanical treatment of atomic structure the book strongly emphasizes the experimental basis of the subject especially in the later chapters It includes ample tutorial material examples illustrations chapter summaries graded problem sets **Pamphlets Articles about Randolph Greenfield**

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