

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of a given acid or hydrogen ion concentration.
2. Calculate hydrogen ion concentrations from pH and hydroxide ion concentrations from pOH.
3. Calculate the pH of weak acids.

Vocabulary

- | | | |
|-------------------|-------------------------------|-----------------------|
| 1. autoionization | 2. hydrogen ion concentration | 3. acid solution |
| 4. autoionization | 5. acid rain | 6. acid rain solution |
| | 7. acid solution | 8. pH |

Key Equation

- | | |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

Part A Completion

Write the hydrogen ion concentration or autoionization constant, if the missing quantity that is indicated in the partial equation below. Complete each equation as needed.

1. When autoionization is 1.0×10^{-14} , autoionization constant is 1.0×10^{-14} .
2. The autoionization constant is 1.0×10^{-14} .
3. The autoionization constant is 1.0×10^{-14} .
4. The autoionization constant is 1.0×10^{-14} .
5. The autoionization constant is 1.0×10^{-14} .
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16. The autoionization constant is 1.0×10^{-14} .
17. The autoionization constant is 1.0×10^{-14} .
18. The autoionization constant is 1.0×10^{-14} .
19. The autoionization constant is 1.0×10^{-14} .
20. The autoionization constant is 1.0×10^{-14} .

1. 1.0×10^{-14}
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3. 1.0×10^{-14}
4. 1.0×10^{-14}
5. 1.0×10^{-14}
6. 1.0×10^{-14}
7. 1.0×10^{-14}
8. 1.0×10^{-14}
9. 1.0×10^{-14}
10. 1.0×10^{-14}
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20. 1.0×10^{-14}

19 2 Hydrogen Ions And Acidity Worksheet Answers

Advisory Council on College Chemistry

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