

## Chemistry Final Exam Question and Answer

which type of reaction does this diagram represent?

Answer: nuclear fusion because atomic nuclei combine to form heavier nucleus.

$4/2 \text{ He} + 4/2 \text{ He} \rightarrow ?$

if this equation was completed, which statement would it best support?

Answer: nuclear fusion produces elements that are heavier than helium.

a fusion reaction takes place between carbon and another element. neutrons are released, and a different element is formed. the different element is?

Answer: heavier than helium.

which best describes a chain reaction associated with a nuclear reaction?

Answer: neutrons released during a fission reaction causes other nuclei to split.

which radioisotope is used as a tracer in nuclear scanning?

Answer: technetium-99m

uranium-232 has a half life of 68.8 years. after 344.0 years how much uranium will remain from a 100.0-g sample?

Answer: 3.13 g

a sample contains 16.75 g of radioisotopes U-236 and 50.25 g of its daughter isotope, Th-232. how long did it take for decay to take place if one half life of U-236 is 23 million years?

Answer: 46 million years

an article about half lives describes a daughter isotope. what is a daughter isotope?

Answer: the isotope that forms from the radioactive decay of a less stable isotope

a radioactive nuclide has 53 protons and 78 neutrons. which is the symbol of this nuclide?

Answer:  $^{131}_{53}\text{I}$

which represents a balanced nuclear equation?

Answer:  $^{24}_{11}\text{Na} \rightarrow ^{24}_{12}\text{Mg} + 0/-1 \text{ e}$

consider the nuclear equation below.

$^{239}_{94}\text{Pu} \rightarrow \text{X} + ^{4}_{2}\text{He}$

what is X?

Answer:  $^{235}_{92}\text{U}$

the symbol for a radioactive nuclide is  $^{210}_{84}\text{Po}$ . which statement is correct?

Answer: the number of neutrons present in the nucleus of each atom is 126.

examine the nuclear reaction  $1/1 \text{ H} + 1/0 \text{ n} \rightarrow 2/1 \text{ H}$ . why is this classified as a nuclear reaction rather than a chemical reaction?

Answer: a change has occurred in a nucleus.

# 70 Question Chemistry Final Exam Answers

**Richard H. Langley, John Moore**



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## **Table of Contents 70 Question Chemistry Final Exam Answers**

1. Understanding the eBook 70 Question Chemistry Final Exam Answers
  - The Rise of Digital Reading 70 Question Chemistry Final Exam Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying 70 Question Chemistry Final Exam Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 70 Question Chemistry Final Exam Answers
  - User-Friendly Interface

4. Exploring eBook Recommendations from 70 Question Chemistry Final Exam Answers
  - Personalized Recommendations
  - 70 Question Chemistry Final Exam Answers User Reviews and Ratings
  - 70 Question Chemistry Final Exam Answers and Bestseller Lists
5. Accessing 70 Question Chemistry Final Exam Answers Free and Paid eBooks
  - 70 Question Chemistry Final Exam Answers Public Domain eBooks
  - 70 Question Chemistry Final Exam Answers eBook Subscription Services
  - 70 Question Chemistry Final Exam Answers Budget-Friendly Options
6. Navigating 70 Question Chemistry Final Exam Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - 70 Question Chemistry Final Exam Answers Compatibility with Devices
  - 70 Question Chemistry Final Exam Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 70 Question Chemistry Final Exam Answers
  - Highlighting and Note-Taking 70 Question Chemistry Final Exam Answers
  - Interactive Elements 70 Question Chemistry Final Exam Answers
8. Staying Engaged with 70 Question Chemistry Final Exam Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 70 Question Chemistry Final Exam Answers
9. Balancing eBooks and Physical Books 70 Question Chemistry Final Exam Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 70 Question Chemistry Final Exam Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 70 Question Chemistry Final Exam Answers
  - Setting Reading Goals 70 Question Chemistry Final Exam Answers
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of 70 Question Chemistry Final Exam Answers
  - Fact-Checking eBook Content of 70 Question Chemistry Final Exam Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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