

- d. Nope, you will have to ask our petty cash custodian for your reimbursement. All disbursements below \$5,000 will have to be paid out of the petty cash fund.
8. By requiring that all deposits must be deposited intact and all disbursements, except petty disbursements, be made through check, you are implementing an internal control called
- Regular bank reconciliations.
 - Imprest system.
 - Voucher system.
 - Bossy system.
9. By requiring that all disbursements must be supported by CDVs, you are implementing an internal control called
- Regular bank reconciliations.
 - Imprest system.
 - Voucher system.
 - Bossy system.
10. Which of the following is incorrect regarding the petty cash fund?
- The petty cash fund should only be used for recurring expenses that are not material enough to warrant disbursement through check.
 - Each petty cash disbursement must be supported by a properly authorized petty cash voucher.
 - The requesting party should affix his or her signature on the petty cash voucher acknowledging the receipt of the petty cash.
 - For a big company, each petty cash voucher must be authorized by the Chief Executive Officer or President.

Chapter 9

Accounting Practice Set

Learning Competencies

The learners should be able to.....

- Perform the steps in the accounting cycle, from preparation of documents to the preparation, analysis, and interpretation of financial statements

Practice Set

The November 30, 20x1 trial balance of Rak N' Koll Co., a call center business, shows the following information:

RAK N' KOLL

Trial Balance
November 30, 20x1

ACCOUNTS	Debit	Credit
Cash	340,000	
Accounts receivable		
Prepaid insurance		
Computer equipment	1,400,000	
Accumulated depreciation		240,000
Loans payable		1,000,000
Owner's equity		1,254,000
Owner's drawings	3,300,000	
Service fees		4,890,000
Salaries expense	990,000	
Rent expense	1,100,000	
Utilities expense	165,000	
Office supplies expense	84,000	
Insurance expense	5,000	
Totals	7,384,000	7,384,000

Accounting Practice Set Answers Tee Time Merchandise

Edith E. Gordon



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web equilibrium moles $hscn$ $2.00 \times 10^{-5} \text{ mol}$ $3.00 \times 10^{-6} \text{ mol}$ $1.70 \times 10^{-5} \text{ mol}$ $hscn$ knowing the number of moles of Fe^{3+} and $hscn$ present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species $mol\ Fe^{3+}$ $1.70 \times 10^{-5} \text{ mol}$ Fe^{3+} $8.50 \times 10^{-4} \text{ M}$

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web equilibrium questions a covalent bond has a standard gibbs free energy of 400 kJ/mol while the hydrogen bond and van der Waals forces have substantially lower values respectively 25 and 5 kJ/mol given such values which of the following is equivalent to the ratio of the K_e value of a covalent bond to that of a hydrogen bond

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web jun 16 2020 summary skills to develop to understand what is meant by chemical equilibrium two methods to describe the likely result of a chemical reaction in the previous chapter we discussed the principles of chemical kinetics which deal with the rate of change or how quickly a given chemical reaction occurs

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web nov 13 2022 define the equilibrium state of a chemical reaction system what is its practical significance state the meaning and significance of the following terms reversible reaction quantitative reaction kinetically inhibited reaction explain the meaning of the statement equilibrium is macroscopically static but microscopically dynamic very

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web in a chemical equilibrium the forward and reverse reactions occur at equal rates and the concentrations of products and reactants remain constant if we run a reaction in a closed system so that the products cannot escape we often find the reaction does not give a 100% yield of products