

**GRADE 10 PHYSICAL SCIENCES**

**NOVEMBER EXAMINATION PAPER 2: CHEMISTRY**

**SECTION A  
QUESTION 1**

Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D ketchup

1.2 Element with the electron configuration:  $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B Na<sub>2</sub>S
- C NaS<sub>2</sub>
- D Na<sub>2</sub>S<sub>2</sub>

1.4 Strongest intermolecular force found in water?

- A Ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a:



- A noble gas
- B halogen
- C alkali earth-metal
- D alkali metal

# Question Paper 2 November 2014 Physical Science

**L Darling-Hammond**



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## **Table of Contents Question Paper 2 November 2014 Physical Science**

1. Understanding the eBook Question Paper 2 November 2014 Physical Science
  - The Rise of Digital Reading Question Paper 2 November 2014 Physical Science
  - Advantages of eBooks Over Traditional Books
2. Identifying Question Paper 2 November 2014 Physical Science
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Question Paper 2 November 2014 Physical Science
  - User-Friendly Interface
4. Exploring eBook Recommendations from Question Paper 2 November 2014 Physical Science
  - Personalized Recommendations
  - Question Paper 2 November 2014 Physical Science User Reviews and Ratings
  - Question Paper 2 November 2014 Physical Science and Bestseller Lists

5. Accessing Question Paper 2 November 2014 Physical Science Free and Paid eBooks
  - Question Paper 2 November 2014 Physical Science Public Domain eBooks
  - Question Paper 2 November 2014 Physical Science eBook Subscription Services
  - Question Paper 2 November 2014 Physical Science Budget-Friendly Options
6. Navigating Question Paper 2 November 2014 Physical Science eBook Formats
  - ePub, PDF, MOBI, and More
  - Question Paper 2 November 2014 Physical Science Compatibility with Devices
  - Question Paper 2 November 2014 Physical Science Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Question Paper 2 November 2014 Physical Science
  - Highlighting and Note-Taking Question Paper 2 November 2014 Physical Science
  - Interactive Elements Question Paper 2 November 2014 Physical Science
8. Staying Engaged with Question Paper 2 November 2014 Physical Science
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Question Paper 2 November 2014 Physical Science
9. Balancing eBooks and Physical Books Question Paper 2 November 2014 Physical Science
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Question Paper 2 November 2014 Physical Science
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Question Paper 2 November 2014 Physical Science
  - Setting Reading Goals Question Paper 2 November 2014 Physical Science
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Question Paper 2 November 2014 Physical Science
  - Fact-Checking eBook Content of Question Paper 2 November 2014 Physical Science
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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