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web calculate the average atomic mass of beanium to be placed on the periodic table beanium isotope lab class set data total mass atoms total mass isotope 1 atoms 1 avg mass 1 total mass isotope 2 atoms 2 avg mass 2 total mass isotope 3 atoms 3 avg

mass 3 total mass of all atoms total of atoms

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web isotopes and weighted average atomic mass amu atomic mass unit mass of one proton mass of one neutron a sample of cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass  $132 \times 0.200 + 133 \times 0.752 + 134 \times 0.047 = 132.85$  amu

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c  $\times 100$  section 3 calculate the average atomic mass of beanium

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web in the beanium lab students obtained a sample of beanium a fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web procedure obtain a sample of the classroom mixture of beanium sort your beanium into its three isotopic bean types count the number of beans in each pile find the mass of each pile of beans determine the average mass of each type of bean based on the samples masses

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web terms in this set 14 what are the objective of the lab calculate from experimental data the atomic mass of the fictitious

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