

Strengths of Acids and Bases

Strengths of Acids

- Strong acid – acid that ionizes completely

- Produce the maximum number of ions
- Good conductors of electricity
- Ex: hydrochloric acid (HCl)

- Weak acid – acid that ionizes only partially

- Some of the acid remains in molecular form
- Ex: acetic acid (vinegar)

- Using the Bronsted-Lowry model, if the reactant acid is strong, its conjugate base is weak

19 3 Chemistry Answers Strengths Of Acids And Bases

S.K. Bhasin, Reena Gupta



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The Pearson Guide to Objective Chemistry for the AIEEE Singhal Atul, 2010-09

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