

0.4 Introduction

The Peugeot 405 model range was introduced into the UK in January 1988 in Saloon form only.

Available with 1.6, 1.8, 1.9 and 2.0 engines, all models have front-wheel-drive with all round independent suspension.

Automatic transmission models were introduced in April 1988.

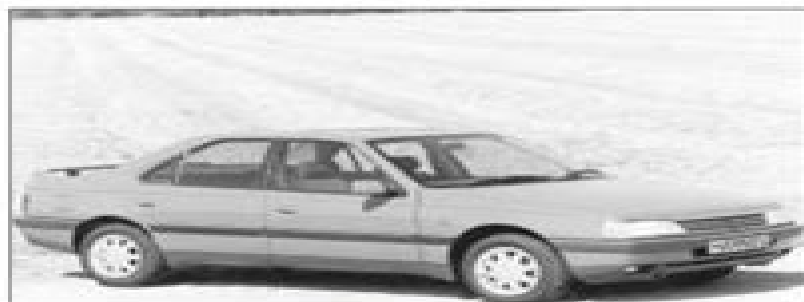
In July 1988 came the sporty Mi 16 version with its 1.9 litre double overhead cam, 16-valve engine, uprated gearbox, suspension and an ABS braking system to match its power.

Estate car versions were introduced in October 1988.

From 1991, engines equipped with catalytic converters were progressively introduced, to meet the more stringent exhaust gas emission regulations.

Since its introduction, the 405 range has continually been developed. All models have a high trim level, which is very comprehensive in the upper model range.

For the home mechanic, the Peugeot 405 is a straightforward vehicle to maintain and repair since design features have been incorporated to reduce the actual cost of ownership to a minimum, and most of the items requiring frequent attention are easily accessible.



Peugeot 405 SRi Saloon



Peugeot 405 GL Estate

The Peugeot 405 Team

Haynes manuals are produced by dedicated and enthusiastic people working in close co-operation. The team responsible for the creation of this book included:

Authors	Steve Rendle Andy Legg
Sub-editor	Carole Turk
Editor & Page Make-up	Bob Jex
Workshop manager	Paul Buckland
Photo Scans	John Martin Paul Tanswell
Cover illustration & Line Art	Roger Hoaling
Wiring diagrams	Matthew Marke

We hope the book will help you to get the maximum enjoyment from your car. By carrying out routine maintenance as described you will ensure your car's reliability and preserve its resale value.

Your Peugeot 405 Manual

The aim of this manual is to help you get the best value from your vehicle. It can do so in several ways. It can help you decide what work must be done (even should you choose to get it done by a garage), provide information on routine maintenance and servicing, and give a logical course of action and diagnosis when random faults occur. However, it is hoped that you will use the manual by tackling the work yourself. On simpler jobs, it may even be quicker than booking the car into a garage and going there twice, to leave and collect it. Perhaps most important, a lot of money can be saved by avoiding the costs a garage must charge to cover its labour and overheads.

The manual has drawings and descriptions to show the function of the various components, so that their layout can be understood. Then the tasks are described and photographed in a clear step-by-step sequence.

Acknowledgements

Thanks are due to Champion Spark Plug who supplied the illustrations showing spark plug conditions. Certain other illustrations are the copyright of the Peugeot Talbot Motor Company Limited, and are used with their permission. Special thanks to Giddons of Taunton who provided several of the project vehicles used in the origination of this manual. Thanks are also due to Sykes-Pickavant Limited, who provided some of the workshop tools, and to all those people at Sparkford who helped in the production of this manual.

We take great pride in the accuracy of information given in this manual, but vehicle manufacturers make alterations and design changes during the production run of a particular vehicle of which they do not inform us. No liability can be accepted by the authors or publishers for loss, damage or injury caused by any errors in, or omissions from, the information given.

Project vehicles

The vehicles used in the preparation of this manual, and which appear in many of the photographic sequences, were a Peugeot 405 GL Saloon, a Peugeot 405 GTX Estate, a Peugeot 405 GR Saloon, and a Peugeot GTX Saloon.

Peugeot 405 Service Guide

RM Cervero



Peugeot 405 Service Guide:

Peugeot 405 (diesel) Service and Repair Manual Steve Rendle,1996 **Peugeot 405 (petrol)** Steve Rendle,A. K. Legg,1996 **Peugeot 405 (Diesel) Service & Repair Manual** Steve Rendle,2001 Saloon Estate inc special limited editions Diesel 1 8 1769cc Turbo 1 9 1905cc Turbo normally aspirated versions *Peugeot 405 (diesel) Service and Repair Manual* Steve Rendle,1996 *Peugeot 405 Owners Workshop Manual* Colin Brown,1990 Peugeot 405 Owner's Workshop Manual Colin Brown,1993 **The Complete Small Truck Cost Guide, 1990** Inc Intellichoice,IntelliChoice, Incorporated,1990-03 The Complete Car Cost Guide, 1991 Peter Levy,1991-03 *Whitaker's Books in Print* ,1998

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Cars & Parts ,1977 *Old Car Value Guide* ,1977 **Autocar & Motor** ,1993-07 Legal Compilation; Statutes and Legislative History, Executive Orders, Regulations, Guidelines and Reports ,1973

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655

furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH

reaction with a 2nd substance $\text{H}_2\text{CO}_3 \text{ aq} + 2\text{NaOH} \text{ aq} \rightarrow \text{Na}_2\text{CO}_3 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction $\text{NaOH} \text{ aq} + \text{C}_6\text{H}_4\text{COOH} \text{ aq} \rightarrow \text{C}_6\text{H}_4\text{COONa} \text{ aq} + \text{H}_2\text{O} \text{ l}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the

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