

DESCRIBING CHEMICAL REACTIONS

Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation

Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical ____1____. The substances that undergo a chemical change are the ____2____. The new substances formed in a chemical reaction are the ____3____. In accordance with the law of conservation of ____4____, a chemical equation must be balanced. When balancing an equation, you place ____5____ in front of reactants and products so that the same number of atoms of each ____6____ are on each side of the equation. An equation must never be balanced by changing the ____7____ in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is ____8____; for a solid, ____9____; and for a gas, ____10____. A substance dissolved in water is designated ____11____. If a ____12____ is used to increase the rate of a chemical reaction, its formula is written above the arrow.

- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- _____ 14. The symbol Δ placed over the arrow in an equation means that heat is supplied to the reaction.
- _____ 15. Atoms are destroyed in a chemical reaction.
- _____ 16. A skeleton equation is not a balanced equation.

Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I
- Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.
 - Acetylene gas (C_2H_2) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

Describing Chemical Change Packet Answers

Ensheng Dong



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Interactive Science For Inquiring Minds Volume B Theory Workbook Express/Normal (Academic) , X-kit FET Grade 12 PHYS SCIENCE CHEMISTRY Clive Lon,2008 *Stride Ahead with Science* □ 6 Madhubun, 1 It is designed in accordance with the latest guidelines laid by NCERT for classes 1 to 8 2 Aims to inculcate inquisitiveness and passion for learning 3 The chapters are designed in a manner that leads to comprehensive learning of concepts development of investigative and scientific skills and the ability to probe into problems and find a possible solution 4 The content of the series is supported by alluring illustrations and attractive layout to lend to the visual appeal and also to enhance the learning experience 5 A clear comprehensive list of learning objectives at the beginning of each chapter 6 A Kick off activity at the beginning of each chapter to set the pace for learning 7 Hand on activities presented using the scientific methodology of having a clear aim and materials required along with recording and discussing the task at hand 8 A section on In Real Life at the end of each chapter imparts value education and helps the learners become a better citizen 9 Evaluation tools in the form of test papers and model test papers in classes 1 to 5 and periodic assessments half yearly paper and a yearly paper in classes 6 to 8

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