

## Physical VS Chemical Changes

Directions: Identify which examples are physical or chemical changes.  
If it's physical, use a 'P'. If it's chemical, use a 'C'.

- P 1. A pencil breaking in half.
- C 2. Iron turning (oxidizing) into rust.
- C 3. Mixing baking soda and vinegar to cause the bubbling and fizzing.
- P 4. Folding clothes after they come out of the dryer.
- C 5. When wood burns and you smell smoke.
- P 6. Clipping your fingernails.
- P 7. Freezing water.
- C 8. When gasoline in an engine combusts (burns) to create exhaust.
- P 9. Changing the shape of a piece of Play-Doh.
- C 10. Acid cooking a raw egg.
- C 11. The leaves of a tree changes from green to brown in the Fall.
- P 12. Smashing a bug.
- C 13. When milk clumps up from spoiling.
- P 14. A balloon popping.
- P 15. Mixing sugar and water.
- C 16. When the food that you eat digests.
- C 17. Combining Mentos and Diet Coke to separate the CO<sub>2</sub>.
- P 18. A paper towel absorbing water.
- C 19. Baking cookies in an oven.
- P 20. Wadding up a piece of paper.

### REMEMBER...

A **physical change** is a usually reversible change in the physical properties of matter, such as size or shape, but does not change the matter itself.

A **chemical change** is an irreversible change in one or more substances into entirely new substances with different properties. The *change* occurring from one substance to another is the **chemical reaction**.

# Describing Chemical Change Answer Key

**Clemens Wendtner**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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