

Chemical Reactions

Section 9.1 Reactions and Equations

pages 282–288

Practice Problems

pages 284–287

Write skeleton equations for the following word equations.

1. Hydrogen and bromine gases react to yield hydrogen bromide.

hydrogen(g) + bromine(g) →
hydrogen bromide(g)



2. When carbon monoxide and oxygen react, carbon dioxide forms.

carbon monoxide(g) + oxygen(g) →
carbon dioxide(g)



3. **Challenge** Write the word equation and the skeleton equation for the following reaction: when heated, solid potassium chlorate yields solid potassium chloride and oxygen gas.

Potassium chlorate(s) →
potassium chloride(s) + oxygen(g)



Write chemical equations for each of the following reactions.

4. In water, iron(III) chloride reacts with sodium hydroxide, producing solid iron(III) hydroxide and sodium chloride.



5. Liquid carbon disulfide reacts with oxygen gas, producing carbon dioxide gas and sulfur dioxide gas.



6. **Challenge** A piece of zinc metal is added to a solution of hydrogen sulfate. This reaction produces a gas and a solution of zinc sulfate.



Section 9.1 Assessment

page 288

7. **Explain** why it is important that a chemical equation be balanced.

Because mass is neither created nor destroyed in chemical reactions, the numbers of atoms of all elements must be equal on both sides of the reaction arrow.

8. **List** three types of physical evidence that indicate a chemical reaction has occurred.

Answers may include release or absorption of energy, change in color, change in odor, formation of a gas, or formation of a solid.

9. **Compare and contrast** a skeleton equation and a chemical equation.

The skeleton equation includes the formulas of reactants and products. The chemical equation gives the relative amounts of reactants and products.

10. **Explain** why it is important to reduce coefficients in a balanced equation to the lowest possible whole-number ratio.

Coefficients in the lowest ratio most clearly indicate the relative amounts of substances in a reaction.

11. **Analyze** When balancing a chemical equation, can you adjust the subscript in a formula? Explain.

No. Doing so changes the identity of the substance.

Describing Chemical Reactions Answer

Arthur Wagner



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Interactive Science For Inquiring Minds Volume B Theory Workbook Express/Normal (Academic) , General Chemistry Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value

DNA Computing and Molecular Programming Luca Cardelli, William Shih, 2011-09 This book constitutes the refereed proceedings of the 17th International Conference on DNA Computing and Molecular Programming DNA17 held in Pasadena CA USA in September 2011 The 12 revised full papers presented together with 5 invited talks were carefully selected from numerous submissions Research in DNA computing and molecular programming draws together mathematics computer science physics chemistry biology and nanotechnology to address the analysis design and synthesis of information based molecular systems This annual meeting is the premier forum where scientists with diverse backgrounds come together with the common purpose of advancing the engineering and science of biology and chemistry from the point of view of computer science physics and mathematics

Chemistry Therald Moeller, 2012-12-02 Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis This book reviews the chemistry of inorganic substances as the science of matter the units of measure used atoms atomic structure thermochemistry nuclear chemistry molecules and ions in action This text also describes the chemical bonds the representative elements the changes of state water and the hydrosphere which also covers water pollution and water purification Water purification occurs in nature through the usual water cycle and by the action of microorganisms The air flushes dissolved gases and volatile pollutants when water seeps through the soil it filters solids as they settle in the bottom of placid lakes Microorganisms break down large organic molecules containing mostly carbon hydrogen nitrogen oxygen sulfur or phosphorus into harmless molecules and ions This text notes that natural purification occurs if the level of contaminants is not so excessive This textbook is suitable for both chemistry teachers and students

Organic Chemistry Douglas C. Neckers, Michael P. Doyle, 1977

Pre-GED Science Arthur Wagner, 1999-07-20 Comprehensive preparation for the Science of the GED Covers thoroughly the areas of biology chemistry earth science and physics It guides students in acquiring such skills as finding main ideas making inferences summarizing information and recognizing cause and effect relationships It also teaches students how to interpret scientific data and find the meaning of scientific words from context

Chemical Misconceptions Keith Taber, 2002 Part 2 provides strategies for dealing with some of the misconceptions that students have by including ready to use classroom resources

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Legislature. Senate,1898

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United States. Patent Office,1890

Theoretical Chemistry Accounts Christopher Cramer,D.G. Truhlar,2013-11-11 For

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Geography of Elections Peter J. Taylor, Ron Johnston, 2014-10-03. Within an international framework, this work provides a fully comprehensive approach to the geographical coverage of elections. Numerous applications of ideas and concepts from human geography are incorporated into a new political context, illustrating the manner in which electoral patterns reflect and help produce the overall geography of a region or state. Discussions of various topics are well supported by numerous maps and diagrams which help clarify arguments and serve to define elections within their basic geographical context.

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