

# DESCRIBING CHEMICAL REACTIONS

## Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation

## Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

## Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical \_\_\_\_1\_\_\_\_. The substances that undergo a chemical change are the \_\_\_\_2\_\_\_\_. The new substances formed in a chemical reaction are the \_\_\_\_3\_\_\_\_. In accordance with the law of conservation of \_\_\_\_4\_\_\_\_, a chemical equation must be balanced. When balancing an equation, you place \_\_\_\_5\_\_\_\_ in front of reactants and products so that the same number of atoms of each \_\_\_\_6\_\_\_\_ are on each side of the equation. An equation must never be balanced by changing the \_\_\_\_7\_\_\_\_ in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is \_\_\_\_8\_\_\_\_; for a solid, \_\_\_\_9\_\_\_\_; and for a gas, \_\_\_\_10\_\_\_\_. A substance dissolved in water is designated \_\_\_\_11\_\_\_\_. If a \_\_\_\_12\_\_\_\_ is used to increase the rate of a chemical reaction, its formula is written above the arrow.

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## Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- \_\_\_\_\_ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- \_\_\_\_\_ 14. The symbol  $\Delta$  placed over the arrow in an equation means that heat is supplied to the reaction.
- \_\_\_\_\_ 15. Atoms are destroyed in a chemical reaction.
- \_\_\_\_\_ 16. A skeleton equation is not a balanced equation.

## Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions.
  - a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.
  - b. Acetylene gas ( $C_2H_2$ ) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

# Describing Chemical Reactions Answers

**R Bogdan**



## **Describing Chemical Reactions Answers:**

## Decoding **Describing Chemical Reactions Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Describing Chemical Reactions Answers**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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