

DETERMINING MOLECULAR FORMULAS (TRUE FORMULAS)

Name Max Lundy

Solve the problems below.

1. The empirical formula of a compound is NO_2 . Its molecular mass is 92 g/mol. What is its molecular formula?

$$\begin{aligned} &14 + 2(16) \\ &= 46 \\ &92 \div 46 = 2 \\ &2(\text{N}) 2(\text{O}_2) \\ &\text{N}_2\text{O}_4 \end{aligned}$$

N_2O_4

2. The empirical formula of a compound is CH_2 . Its molecular mass is 70 g/mol. What is its molecular formula?

$$\begin{aligned} &12 + 2 \\ &= 14 \\ &70 \div 14 = 5 \\ &5(\text{C}) 5(\text{H}_2) \\ &\text{C}_5\text{H}_{10} \end{aligned}$$

C_5H_{10}

3. A compound is found to be 40.0% carbon, 6.7% hydrogen and 53.3% oxygen. Its molecular mass is 60 g/mol. What is its molecular formula?

$$\begin{aligned} &\text{(C)} \quad 40 \left(\frac{1}{12} \right) = 3.333 \quad \frac{60}{12} = 5 \quad \text{C}_5\text{H}_{10}\text{O} \\ &\text{(H)} \quad 6.7 \left(\frac{1}{1} \right) = 6.7 \quad \frac{60}{1} = 60 \quad 2(\text{C}) 2(\text{H}_2) 2(\text{O}) \\ &\text{(O)} \quad 53.3 \left(\frac{1}{16} \right) = 3.333 \quad \frac{60}{16} = 3.75 \quad \text{C}_2\text{H}_4\text{O}_2 \end{aligned}$$

$\text{C}_2\text{H}_4\text{O}_2$

4. A compound is 64.9% carbon, 13.5% hydrogen and 21.6% oxygen. Its molecular mass is 74 g/mol. What is its molecular formula?

$$\begin{aligned} &\text{(C)} \quad 64.9 \left(\frac{1}{12} \right) = 5.4083 \quad \frac{74}{12} = 6.166 \\ &\text{(H)} \quad 13.5 \left(\frac{1}{1} \right) = 13.5 \quad \frac{74}{1} = 74 \\ &\text{(O)} \quad 21.6 \left(\frac{1}{16} \right) = 1.35 \quad \frac{74}{16} = 4.625 \end{aligned}$$

$\text{C}_6\text{H}_{10}\text{O}$

5. A compound is 54.5% carbon, 9.1% hydrogen and 36.4% oxygen. Its molecular mass is 88 g/mol. What is its molecular formula?

$$\begin{aligned} &\text{(C)} \quad 54.5 \left(\frac{1}{12} \right) = 4.5416 \quad \frac{88}{12} = 7.333 \\ &\text{(H)} \quad 9.1 \left(\frac{1}{1} \right) = 9.1 \quad \frac{88}{1} = 88 \\ &\text{(O)} \quad 36.4 \left(\frac{1}{16} \right) = 2.275 \quad \frac{88}{16} = 5.5 \end{aligned}$$

$\text{C}_7\text{H}_{12}\text{O}_2$

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Determining Molecular Formulas True Formulas Answer Key

JG Myers



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web the internal parts of female sexual anatomy or what's typically referred to as female include vagina the vagina is a tube

that connects your vulva with your cervix and uterus it's what babies and menstrual blood leave the body through it's also where some people put penises fingers sex toys menstrual cups and or tampons

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