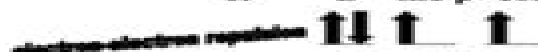


Chemistry diagnostic test (Topic 12)

1. B – the  $p^4$  config. of S is less stable, electron- electron repulsion.



2. A – B is lower than Be because two s electrons shield the 2p electrons
3. E – Au and Ag both have full d orbitals(10electrons) ( $s^1$ )
4. E – electronegativity increases from lower left to upper right (F-most electronegative)
5. B – sulfur's oxidation numbers: 2-, 2+, 4+, 6+
6. A – atomic radii increase right to left and top to bottom
7. C – cations are smaller, the higher the charge the smaller the cation
8. A – Group 1 has low 1<sup>st</sup> ionization energy, but 2<sup>nd</sup> is pulling an inner core electron(big jump)
9. E – equal moles - 0.25
10. D – greatest molar mass
11. D – greatest mass/volume
12. E – all same temp
13. E – Phosphorus – high reactivity with oxygen not found free in nature
14. B – not all have a 2+ oxidation
15. A – all substances contain London dispersion forces
16. D – hydrogen must be bonded to nitrogen, oxygen or fluorine
17. A – propane (molar mass – 44) dispersion forces only – lowest temp  
Ethanol (molar mass – 46) also has hydrogen bonding  
Water (molar mass – 18) very strong hydrogen bonds
18. C
19. A – balanced equation is  
 $\text{Na}_2\text{CO}_3 + 2\text{HCl} \rightarrow 2\text{NaCl} + \text{CO}_2 + \text{H}_2\text{O}$   
Limiting is sodium carbonate
20. C –  $(0.300 \text{ moles/L})(.300\text{L}) = 0.0900 \text{ mol Cl}^- \text{ ion} - \text{requires } 1/3 \text{ as much aluminum chloride or } 0.0300 \text{ mol}$   
 $0.0300 \text{ mol}/0.500\text{M/mol/L} = 0.0600\text{L} = 60.0 \text{ ml}$
21. B -  $0.500 \text{ M} \times 0.250\text{L} \times 200\text{g/mol} = 25.0 \text{ g}$
22. D-  $2 \text{ C}_3\text{H}_8\text{O} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 8\text{H}_2\text{O}$
23. C – 3.00 carbon =  $\frac{1}{2}$  mole of C therefore  $\frac{1}{2}$  mol of Z combines  
 $0.5 \text{ mol Z} \times 36.0 \text{ g/mol}(3\text{timesC}) = 18.0 \text{ g Z}$
24. E – 4 moles of O
25. C – an alkali metal reacts with water to produce hydrogen gas and an aqueous alkali metal hydroxide
26. D –  $\text{H}_2\text{SO}_4 + \text{NaOH} \rightarrow \text{NaHSO}_4 + \text{H}_2\text{O}$  (diprotic acid and monobasic base)
27. E – X has a 3+ charge
28. E
29. B – 1 mole of magnesium requires 2 moles of HCl therefore HCl is limiting
30. D – a formation reaction produces 1 mole of a product

# Diagnostic Test Topic 10 Chemistry Answer Key

**Disha Experts**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Disha Experts' text.

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web the pad foundations are shallow foundation that take and spread point loads to the soil safely the pad foundation are preferred if the soil at the site have sufficient strength and is not too deep to construct the thickness of pad foundations are generally uniform

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web 6 8 structural design of foundations 6 6 9 preparation of the subsoil 2 section 6 of en 1997 1 applies to pad strip and raft foundations and some the worked examples in this chapter consider the design of a pad footing on dry sand example 10 1 the same footing but eccentrically loaded example 10 2 a strip footing on clay

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web jan 7 2020 the soil's bearing capacity as well as the concrete's strength are the factors that define the angle of the load's spread the type of skeletal frame will determine the pad foundation design for example a cast-in-place concrete column will require a kicker and continuity bars to be cast into the pad

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web this technical guidance note covers the design of concrete pad foundations both mass and reinforced concrete forms it will not however discuss how the bearing capacity of the soil is determined as that is explained in technical guidance note

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web design of pad footing cracking detailing requirements all reinforcements should extend the full length of the footing if 1 5 3 at least two thirds of the reinforcement parallel to l y should be concentrated in a band width 3 centred at column where l x l y and c x c y are the footing and column dimension in x and y directions

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web design examples 5 and 6 in chapter 11 figures h 2 h 3 and h 4 preliminary estimation of effective depth required for reinforced pad bases in bending these design charts give a preliminary estimate of the effective depth required of a pad foundation in bending beam shear and punching shear respectively given the

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