

Dilutions – Practice Problems

Instructions: Use the equation for dilutions to complete the problems below.

- 1) A student obtains 0.30 liters of 2.0 M silver nitrate solution. She dilutes the solution until it has a volume of 2.6 liters. What is the **molarity** of the new solution?

$$M_1 V_1 = M_2 V_2$$

$$(2.0) (0.30) = M_2 (2.6)$$

$$(0.6) = M_2 (2.6)$$

$$M_2 = 0.23 \text{ M}$$

- 2) Before an experiment, a scientist obtains 2.45 L of 1.25 M sulfuric acid (H_2SO_4) and dilutes it until the molarity drops to 0.80 M. What is the **volume** of the new solution?

$$M_1 V_1 = M_2 V_2$$

$$(1.25) (2.45) = (0.8) V_2$$

$$(3.06) = (0.8) V_2$$

$$V_2 = 3.83 \text{ L}$$

- 3) A solution of hydrochloric acid is diluted, bringing the volume from 2.0 L to 8.5 L. If the solution has a final molarity of 1.4 M, what was the **molarity before** dilution?

$$M_1 V_1 = M_2 V_2$$

$$M_1 (2.0) = (1.4) (8.5)$$

$$(250.0) = (2.0) V_2$$

$$V_2 = 125.0 \text{ mL}$$

- 4) If a student adds 250 mL of water to a 300 mL solution of NaCl that has a concentration of 3.0 M, what will be the **molarity** of the final solution?

$$M_1 V_1 = M_2 V_2$$

$$(3.0) (300) = M_2 (550)$$

$$(900) = M_2 (550)$$

$$M_2 = 1.64 \text{ M}$$



- 5) A scientist adds 1.2 liters of water to a 5.0 liter solution, causing the molarity to become 0.4 M. What was the **molarity** of the **initial solution** before the dilution?

$$M_1 V_1 = M_2 V_2$$

$$M_1 (5.0) = (0.4) (7.2)$$

$$M_1 (5.0) = 2.88$$

$$M_1 = 0.576 \text{ M}$$

Dilution Notes And Practice Answers

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Dilution Notes And Practice Answers:

Fundamental Laboratory Mathematics Lela Buckingham, 2014-05-16 Conquer the math skills essential for the laboratory and reduce the anxieties math often induces Step by step skill by skill you ll progress from simple to complex calculations building your proficiencies and testing them along the way Perfect for classroom clinical and professional success *NEET UG Physics Study Notes with Theory + Practice MCQs for Complete Preparation | Based on New Syllabus as per NMC* EduGorilla Prep Experts, 2023-12-01 **Ebook: Chemistry** Julia Burdge, 2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text *Food Analysis Laboratory Manual* S. Suzanne Nielsen, 2017-06-07 This third edition laboratory manual was written to accompany Food Analysis Fifth Edition by the same author New to this third edition of the laboratory manual are four introductory chapters that complement both the textbook chapters and the laboratory exercises The 24 laboratory exercises in the manual cover 21 of the 35 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component or characteristic Most of the laboratory exercises include the following background reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis *NBS Technical Note* , 1974-08 , **U Can: Chemistry I For Dummies** John T. Moore, Chris Hren, Peter J. Mikulecky, 2015-08-10 Now you can score higher in chemistry Every high school requires a course in chemistry for graduation and many universities require the course for majors in medicine engineering biology and various other sciences U Can Chemistry I For Dummies offers all the how to content you need to enhance your classroom learning simplify complicated topics and deepen your understanding of often intimidating course material Plus you ll find easy to follow examples and hundreds of practice problems as well as access to 1 001 additional Chemistry I practice problems online As more and more students enroll in chemistry courses the need for a trusted and accessible resource to aid in study has never been greater That s where U Can Chemistry I For Dummies comes in If you re struggling in the classroom this hands on friendly guide makes it easy to conquer chemistry Simplifies basic chemistry principles Clearly explains the concepts of matter and energy atoms and molecules and acids and bases Helps you tackle problems you may face in your Chemistry I course Combines how to with try it to form one perfect resource for chemistry students If you re confused by chemistry and want to increase your chances of scoring your very best at exam time U Can Chemistry I For Dummies shows you that you can Nielsen's Food Analysis Laboratory Manual B. Pam Ismail, S. Suzanne Nielsen, 2024-06-20 This fourth edition

laboratory manual was written to accompany Nielsen's Food Analysis Sixth Edition by the same authors. New to this fourth edition of the laboratory manual are three new chapters that complement both the textbook chapters and the laboratory exercises. The book again contains four introductory chapters that help prepare students for doing food analysis laboratory exercises. The 26 laboratory exercises in the manual cover 24 of the 35 chapters in the textbook. Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component or characteristic. Most of the laboratory exercises include the following background reading assignment, objective, principle of method, chemicals, reagents, precautions, and waste disposal, supplies, equipment, procedure, data, and calculations, questions, and references. This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis. Annual Book of ASTM Standards, American Society for Testing and Materials, 2003. A compilation of all ASTM standards issued each year. **Book of ASTM Standards, with Related Material**, American Society for Testing and Materials, 1977. *Chemistry*, Neil D. Jespersen, Alison Hyslop, 2021-11-02. *Chemistry: The Molecular Nature of Matter*, 8th Edition, continues to focus on the intimate relationship that exists between structure at the atomic/molecular level and the observable macroscopic properties of matter. Key revisions in this edition focus on three areas: The deliberate inclusion of more updated real-world examples that relate common real-world student experiences to the science of chemistry. Simultaneously, examples and questions have been updated to align them with career concepts relevant to the environmental, engineering, biological, pharmaceutical, and medical sciences. Providing students with transferable skills with a focus on integrating metacognition and three-dimensional learning into the text. When students know what they know, they are better able to learn and incorporate the material. Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment, answer-specific responses, and additional practice resources. The 8th edition continues to emphasize the importance of applying concepts to problem-solving to achieve high-level learning and increase retention of chemistry knowledge. Problems are arranged in an intuitive confidence-building order. Theory of Chemical Pathology, Ronald A. McAllister, 1967. Basic Laboratory Methods for Biotechnology, Lisa A. Seidman, Cynthia J. Moore, Jeanette Mowery, 2021-12-28. *Basic Laboratory Methods for Biotechnology*, Third Edition, is a versatile textbook that provides students with a solid foundation to pursue employment in the biotech industry and can later serve as a practical reference to ensure success at each stage in their career. The authors focus on basic principles and methods while skillfully including recent innovations and industry trends throughout. Fundamental laboratory skills are emphasized, and boxed content provides step-by-step laboratory method instructions for ease of reference at any point in the student's progress. Worked-through examples and practice problems and solutions assist student comprehension. Coverage includes safety practices and instructions on using common laboratory instruments. Key Features: Provides a valuable reference for laboratory professionals at all stages of their careers. Focuses on basic principles and methods to provide students with the knowledge needed to begin a career in the Biotechnology industry. Describes

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Optimal Design for Nonlinear Response Models Valerii V. Fedorov,Sergei L. Leonov,2013-07-15 Optimal Design for Nonlinear Response Models discusses the theory and applications of model based experimental design with a strong emphasis on biopharmaceutical studies The book draws on the authors many years of experience in academia and the pharmaceutical industry While the focus is on nonlinear models the book begins with an explanation of the key ideas using linear models as examples Applying the linearization in the parameter space it then covers nonlinear models and locally optimal designs as well as minimax optimal on average and Bayesian designs The authors also discuss adaptive designs focusing on procedures with non informative stopping The common goals of experimental design such as reducing costs supporting efficient decision making and gaining maximum information under various constraints are often the same across diverse applied areas Ethical and regulatory aspects play a much more prominent role in biological medical and pharmaceutical research The authors address all of these issues through many examples in the book

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