

MENDELEEV'S PERIODIC TABLE

Introduction

The periodic table of elements arranges all of the known chemical elements in grid-like chart. The modern periodic table organizes elements from left to right in order of increasing atomic number, which is the number of protons in each element's nucleus. Depending on where an element is located on the table, many of its properties can be predicted, including mass, size, and chemical reactivity.

The Periodic Table of Elements

However, the elements weren't always organized in such a clear way. For many years, chemists looked for a logical way to organize the known elements. This became increasingly important as more and more elements were discovered. In 1869, an ingenious method of organizing elements was created by a Russian scientist named **Dmitri Mendeleev**.

WORKSHEET



Mendeleev's Periodic Table

physical and chemical properties. Mendeleev then arranged and rearranged the cards in countless different ways, looking for a relationship to their physical and chemical properties.



Mendeleev began to notice a pattern when he arranged the elements in order of increasing atomic mass. But the pattern was not perfect. Mendeleev worked intensely on his "game" for three straight days, after which he went to sleep. According to Mendeleev, he had a vivid dream in which all of the elements fell into place. When he woke up, he immediately wrote the idea on a piece of paper, which only needed minor corrections as he perfected his theory. A sample of Mendeleev's early periodic table follows on the next page.

Mendeleev's First Periodic Table

From left to right across each row of his table, elements were arranged by increasing atomic mass. Mendeleev discovered that if he placed eight elements in each row and then continued on to a new row, the columns of the table would contain elements with similar properties. He called the

Dmitri Mendeleev Worksheet Answers

Christian Drosten



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